(19)

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European Patent Office

Office européen des brevets



(11) EP 1 349 435 A1

(12)

EUROPEAN PATENT APPLICATION published in accordance with Art. 158(3) EPC

- (43) Date of publication: 01.10.2003 Bulletin 2003/40
- (21) Application number: 01999132.2
- (22) Date of filing: 30.11.2001

- (51) Int CI.7: **H05B 33/14**, C09K 11/06, C07F 15/00, C07F 19/00
- (86) International application number: PCT/JP01/10477
- (87) International publication number: WO 02/045466 (06.06.2002 Gazette 2002/23)
- (84) Designated Contracting States:

 AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU

 MC NL PT SE TR

 Designated Extension States:

 AL LT LV MK RO SI
- (30) Priority: 30.11.2000 JP 2000364350 08.03.2001 JP 2001064203
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(54) LUMINESCENT ELEMENT AND DISPLAY

(57) In a luminescence device formed of one or plural layers of organic film between a cathode and an anode, at least one layer is a luminescence layer, and a luminescence molecule of a metal coordination compound having a basic structure represented by formula (1) below and having a substituent on at least one of cyclic groups A and B is incorporated as a guest in a host material at a concentration of at least 8 wt. %, which

is higher than a concentration at which a luminescence molecule of a similar structure but having no substituent exhibits a maximum luminescence efficiency to form the luminescence layer. As a result, a high-efficiency luminescence device is provided, which is less liable to cause concentration extinction even when a luminescence molecule is contained at a high-concentration relative to the host material in the luminescence layer.

FIG. 1

(b)

11 METAL ELECTRODE

16 ELECTRON-TRANSPORTING LAYER

12 LUMINESCENCE LAYER

13 HOLE-TRANSPORTING LAYER

14 TRANSPARENT ELECTRODE

15 TRANSPARENT SUBSTRATE

Printed by Jouve, 75001 PARIS (FR)

Description

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[TECHNICAL FIELD]

[0001] The present invention relates to an organic film luminescence device for use in a planar light source, a planar display, etc.

[0002] The present invention relates to a luminescence device using an organic compound, more particularly to a high-efficiency luminescence device containing in its luminescence layer a luminescence material-comprising a metal coordination compound less liable to cause concentration extinction even when used at a high-concentration.

[BACKGROUND ART]

[0003] An old example of organic luminescence device is, e.g., one using luminescence of a vacuum-deposited anthracene film (Thin Solid Films, 94 (1982) 171). In recent years, however, in view of advantages, such as easiness of providing a large-area device compared with an inorganic luminescence device, and possibility of realizing desired luminescence colors by development of various new materials and drivability at low voltages, an extensive study thereon for device formation as a luminescence device of a high-speed responsiveness and a high efficiency, has been conducted.

[0004] As described in detail in, e.g., Macromol. Symp. 125, 1 - 48 (1997), an organic EL device generally has a structure comprising upper and lower two electrodes and a plurality of organic film layers between the electrodes formed on a transparent substrate. Basic structures thereof are shown in Figures 1(a) and (b).

[0005] As shown in Figure 1, an organic EL device generally has a structure comprising a transparent, electrode 14, a metal electrode 11, and a plurality of organic film layers therebetween on a transparent substrate 15.

[0006] In the device of Figure 1(a), the organic layers comprise a luminescence layer 12 and a hole-transporting layer 13. For the transparent electrode 14, ITO, etc., having a large work function are used, for providing a good hole-injection characteristic from the transparent electrode 14 to the hole-transporting layer 13. For the metal electrode 11, a metal, such as aluminum, magnesium or an alloy of these, having a small work function is used for providing a good electron-injection characteristic. These electrodes have a thickness of 50 - 200 nm.

[0007] For the luminescence layer 12, aluminum guinolynol complexes (a representative example thereof is Alq3 shown hereinafter), etc., having an electron-transporting characteristic and luminescence characteristic are used. For the hole-transporting layer, biphenyldiamine derivatives (a representative example thereof is α -NPD shown hereinafter), etc., having an electron-donative characteristic are used.

[0008] The above-structured device has a rectifying characteristic, and when an electric field is applied between the metal electrode 11 as a cathode and the transparent electrode 14 as an anode, electrons are injected from the metal electrode 11 into the luminescence layer 12 and holes are injected from the transparent electrode 15. The injected holes and electrons are recombined within the luminescence layer 12 to form excitons and cause luminescence. At this time, the hole-transporting layer 13 functions as an electron-blocking layer to increase the recombination efficiency at a boundary between the luminescence layer 12 and hole-transporting layer 13, thereby increasing the luminescence efficiency.

[0009] Further, in the structure of Figure 1(b), an electron-transporting layer 16 is disposed between the metal electrode 11 and the luminescence layer 12. By separating the luminescence and the electron and hole-transportation to provide a more effective carrier blocking structure, effective luminescence can be performed. For the electron-transporting layer 16, an electron-transporting material, such as an oxidiazole derivative, is used.

[0010] Known luminescence processes used heretofore in organic EL devices include one utilizing an excited singlet state and one utilizing an excited triplet state, and the transition from the former state to the ground state is called "fluorescence" and the transition from the latter state to the ground state is called "phosphorescence". And the substances in these excited states are called a singlet exciton and a triplet exciton, respectively.

[0011] In most of the organic luminescence devices studied heretofore, fluorescence caused by the transition from the excited singlet state to the ground state, has been utilized. On the other hand, in recent years, devices utilizing phosphorescence via triplet excitons have been studied.

[0012] Representative published literature may include:

Article 1: Improved energy transfer in electrophosphorescent device (D:F. O'Brien, et al., Applied Physics Letters, Vol. 74, No. 3, p. 422 (1999)); and

Article 2: Very high-efficiency green organic light-emitting devices based on electrophosphorescence (M.A. Baldo, et al., Applied Physics Letters, Vol. 75, No. 1, p. 4 (1999)).

[0013] In these articles, a structure including 4 organic layers devices as shown in Figure 1(c) has been principally

used, including, from the anode side, a hole-transporting layer 13, a luminescence layer 12, an exciton diffusion-prevention layer 17 and an electron-transporting layer 11. Materials used therein include carrier-transporting materials and phosphorescent materials, of which the names and structures are shown below together with their abbreviations.

Alg3: aluminum quinolinol complex

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α-NPD: N4,N4'-di-naphthalene-1-yl-N4,N4'-diphenyl-biphenyl-4,4'-diamine

CBP: 2,9-dimethyl-4, 7-diphenyl-1,10-phenanthroline

PtOEP: platinum-octaethylporphyrin complex

Ir(ppy)3: iridium-phenylpyrimidine complex

[0014] The above-mentioned Articles 1 and 2 both have reported structures, as exhibiting a high efficiency, including a hole-transporting layer 13 comprising α -NPD, an electron-transporting layer 16 comprising Alq3, an exciton diffusion-preventing layer 17 comprising BCP, and a luminescence layer 12 comprising CBP as a host and ca. 6 % of platinum-octaethylporphyrin complex (PtOEP) or iridium-phenylpyrimidine complex (Ir(ppy)₃) as a phosphorescent material dispersed in mixture therein.

Ir(ppy)3

[0015] Such a phosphorescent material is particularly noted at present because it is expected to provide a high luminescence efficiency in principle for the following reasons. More specifically, excitons formed by carrier recombination comprise singlet excitons and triplet excitons in a probability ratio of 1:3. Conventional organic-EL devices have utilized fluorescence of which the luminescence efficiency is limited to at most-25 %. On the other hand, if phosphorescence generated from triplet excitons is utilized, an efficiency of at least three times is expected, and even an efficiency of 100 %, i.e., four times, can be expected in principle, if a transition owing to intersystem-crossing from a singlet state having a higher energy to a triplet state is taken into account.

[0016] However, like a fluorescent-type device, such an organic luminescence device utilizing phosphorescence is generally required to be further improved regarding the deterioration of luminescence efficiency and device stability.

[0017] The reason of the deterioration has not been fully clarified, but the present inventors consider as follows based on the mechanism of phosphorescence.

[0018] In the case where the luminescence layer comprises a host material having a carrier-transporting function

and a phosphorescent guest material, a process of phosphorescence via triplet excitons may include unit processes as follows:

- 1. transportation of electrons and holes within a luminescence layer,
- 2. formation of host excitons,

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- 3. excitation energy transfer between host molecules,
- 4. excitation energy transfer from the host to the guest,
- 5. formation of guest triplet excitons, and
- 6. transition of the guest triplet excitons to the ground state and phosphorescence.

[0019] Desirable energy transfer in each unit process and luminescence are caused in competition with various energy deactivation processes.

[0020] Particularly, in a phosphorescent material, this may be attributable to a life of the triplet excitons which is longer by three or more digits than the life of a singlet exciton. More specifically, because it is held in a high-energy excited state for a longer period, it is liable to react with surrounding materials and cause polymer formation among the excitons, thus incurring a higher probability of deactivation process resulting in a material change or life deterioration, as we have considered.

[0021] Needless to say, a luminescence efficiency of an organic luminescence device is increased by increasing the luminescence quantum yield of a luminescence center material, but is is also an important factor for enhancing the luminescence intensity of the device to increase the concentration of a luminescence material in the luminescence layer.

[0022] The luminescence intensity is increased in proportion to the concentration of a luminescence material in a luminescence layer in the case of a low concentration (up to several wt. %) of the luminescence material in the luminescence layer. However, above several % or 7 %, a deviation from the proportional relationship is observed, and the luminescence intensity is rather lowered to result in a worse efficiency. This phenomenon is reported in Japanese Laid-Open Patent Application (JP-A) 05-078655, JP-A 05-320633, etc., and is known as concentration extinction or concentration deactivation.

[0023] Actually, in the case of using Ir(ppy)₃ in CBP as the host material, the best luminescence efficiency is attained at a concentration of ca. 6 - 7 %, and the luminescence efficiency is rather lowered thereabove, down to about a half at 12 % concentration and 1/10 or below at 100 % concentration (Applied Physics Letters 4, vol. 75, 1999).

[0024] The phenomenon is caused by abundant presence of molecules in the triplet excited state waiting for luminescence in the case of a phosphorescence substance having a life of triplet exciton longer by 3 digits or more than the life of singlet exciton. In this state, thermal deactivation of losing energy due to mutual interaction of triplet excitons is liable to occur. This is called triplet-triplet extinction and is associated with a lowering in luminescence efficiency at a high current density. Further, it is also considered that due to a long retention time at a high energy state, the excitons are caused to have an increased probability of reaction with a surrounding material and formation of polymers of excitons causing deactivation, or even leading to material change or deterioration of life.

[DISCLOSURE OF INVENTION]

[0025] An object of the present invention is to provide an organic luminescence device of a higher luminescence intensity by suppressing the above-mentioned concentration extinction phenomenon and providing an environment of using a luminescence material at a higher concentration.

[0026] More specifically, an object of the present invention is to provide a luminescence material less liable to cause concentration extinction even when used at a high concentration with respect to the host material in a luminescence layer by introducing a substituent group into a metal coordination compound as such a luminescence material.

[0027] A more specific object of the present invention is to provide an organic luminescence device capable of a large luminescence intensity, that is an organic luminescence device, comprising: a pair of electrodes each disposed on a substrate, and at least one luminescence layer comprising an organic compound disposed between the electrodes; wherein the luminescence layer comprises a non-luminescent first organic compound and a phosphorescent second organic compound represented by formula (1) shown below, and the second organic compound is present at a concentration of at least 8 wt. % in the luminescence layer:

$$\mathsf{ML}_{\mathsf{m}}\mathsf{L'}_{\mathsf{n}}$$
 (1),

wherein M is a metal atom of Ir, Pt, Rh or Pd; L and L' are mutually different bidentate ligands; m is 1, 2 or 3; n is 0, 1 or 2 with the proviso that m+n is 2 or 3; a partial structure ML_m is represented by formula:(2) shown below and a partial

structure ML'_n is represented by formula (3), (4) or (5) shown below:

$$\begin{bmatrix}
A \\
C \\
N \\
B
\end{bmatrix}$$
(2)
$$\begin{bmatrix}
A \\
C \\
C \\
N \\
B^*
\end{bmatrix}$$
(3)
$$\begin{bmatrix}
A \\
N \\
B^*
\end{bmatrix}$$
(4)
$$\begin{bmatrix}
A \\
O \\
O \\
G
\end{bmatrix}$$
(5)

wherein N and C are nitrogen and carbon atoms, respectively; A and A' are respectively a cyclic group capable of having a substituent and bonded to the metal atom M via the carbon atom; B, B' and B" are respectively a cyclic group represented by a formula of (6) - (14) shown below capable of having a substituent and connected to the metal atom M via the nitrogen atom:

[0028] According to another aspect, the organic luminescence device of the present invention is a luminescence device comprising one or plural layers of organic film disposed between a cathode and an anode and including at least one layer of luminescence layer, that is characterized by containing a luminescence molecule of the formula (1) having a substituent and showing a maximum luminescence characteristic at a concentration higher than a concentration at which a luminescence molecule of a similar structure but having no substituent shows a maximum luminescence efficiency.

[0029] More specifically, in a luminescence device comprising one or plural layers of organic film between a cathode and an anode, including at least one layer of luminescence layer, it is preferred that a luminescence molecule of the formula (1) including at least one cyclic group having a substituent is contained at a concentration higher than a concentration at which a luminescence molecule of a similar structure shows a maximum luminescence efficiency.

[BRIEF DESCRIPTION OF THE DRAWINGS]

[0030]

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Figure 1 illustrates embodiments of the luminescence device according to the present invention.

Figure 2 illustrates a simple matrix-type organic EL device according to Example-28.

Figure 3 illustrates drive signals used in Example 28.

Figure 4 schematically illustrates a panel structure including an EL device and drive means.

Figure 5 illustrates an example of pixel circuit.

Figure 6 is a schematic view showing an example of sectional structure of a FFT substrate.

5 [BEST MODE FOR PRACTICING THE INVENTION]

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[0031] Basic device structures according to the present invention are similar to those shown in Figures 1(a), (b) and (c).

[0032] More specifically, as shown in Figure 1, an organic luminescence device generally-comprises, on a transparent electrode 15, a 50 to 200 nm-thick transparent electrode 14, a plurality of organic film layers and a 10 to 500 nm-thick metal electrode 11 formed so as to sandwich the organic layers.

[0033] Figure 1(a) shows an embodiment wherein the organic luminescence device comprises a luminescence layer 12 and a hole-transporting layer 13. The transparent electrode 14 may comprise ITO, etc., having a large work function so as to facilitate hole injection from the transparent electrode 14 to the hole-transporting layer 13. The metal electrode 11 comprises a metal material having a small work function, such as aluminum, magnesium or alloys of these elements, so as to facilitate electron injection into the organic luminescence device.

[0034] The luminescence layer 12 comprises a compound according to the present invention. The hole-transporting layer 13 may comprise, e.g., a triphenyldiamine derivative, as represented by α -NPD mentioned above, and also a material having an electron-donative property as desired.

20 [0035] A device organized above exhibits a current-rectifying characteristic, and when an electric field is applied between the metal electrode 11 as a cathode and the transparent electrode 14 as an anode, electrons are injected from the metal electrode 11 into the luminescence layer 12, and holes are injected from the transparent electrode 15. The injected holes and electrons are recombined in the luminescence layer 12 to form excitons, which cause luminescence. In this instance, the hole-transporting layer 13 functions as an electron-blocking layer to increase the recombination efficiency at the boundary between the luminescence layer layer 12 and the hole-transporting layer 13, thereby providing an enhanced luminescence efficiency.

[0036] Further, in the structure of Figure 1(b), an electron-transporting layer 16 is disposed between the metal electrode 11 and the luminescence layer 12 in Figure 1(a). As a result, the luminescence function is separated from the functions of election transportation and hole transportation to provide a structure exhibiting more effective carrier blocking, thus increasing the luminescence efficiency. The electron-transporting layer 16, may comprise, e.g., an oxadiazole derivative.

[0037] Figure 1(c) shows another desirable form of a four-layer structure, including a hole-transporting layer 13, a luminescence layer 12, an exciton diffusion prevention layer 17 and an electron-transporting layer 16, successively from the side of the transparent electrode 14 as an anode.

35 [0038] Each of the organic film layers 12, 13, 16 and 17 is formed in a thickness of at most 200 nm, and particularly the luminescence layer 12 is formed in a thickness of 5 - 200 nm.

[0039] The present inventors have got a knowledge that the use of a metal coordination compound including a substituted cyclic group and represented by the above-mentioned formula (1) allows a high efficiency luminescence and provides less liability of concentration extinction even at a high concentration than the conventional level due to suppression of inter-molecular interaction.

[0040] It has been also found that the suppression of concentration extinction is an effect attributable to a substituent possessed by the metal coordination compound, and the concentration extinction becomes less liable to occur not regardless of the coordination number of ligands but owing to the presence of a substituent on at least one ligand.

[0041] Particularly, as a result, in a conventional phosphorescence-type organic-EL device, the luminescence material can be used at a high concentration of 8 % or higher in the luminescence layer, thereby providing an organic-EL device exhibiting a high luminescence luminance.

[0042] The metal coordination compound used in the present invention emits phosphorescence, and its lowest excited state is believed to be an MLCT* (metal-to-ligand charge transfer) excited state or π - π * excited state in a triplet state, and phosphorescence is caused at the time of transition from such a state to the ground state.

[0043] It is generally said that phosphorescence life is shorter at MCLT than at π-π, but the molecular structure suppressing the concentration used in the present invention is effective for both MCLT and π-π as the lowest excited state, and the molecule can be doped at a high concentration in the luminescence layer in either case.

[0044] The luminescence material of the present invention exhibited a high phosphorescence yield of from 0.1 to 0.9 and a short phosphorescence life of 0.1 - 30 µsec. The phosphorescence yield referred to herein is a relative quantum yield, i.e., a ratio of an objective sample's quantum yield \$(sample) to a standard sample's quantum yield \$(sample) to a standard sample squantum yield \$(sample squantum yield \$(sample

(st)) and is determined according to the following formula:

$\bar{\Phi}(\text{sample})/\bar{\Phi}(\text{st}) =$

[Sem(sample)/labs(sample)]/[Sem(st)/labs(st)],

wherein labs(st) denotes an absorption coefficient at an excitation wavelength of the standard sample; Sem(st), a luminescence spectral areal intensity when excited at the same wavelength: labs(sample), an absorption coefficient at an excitation wavelength of an objective compound; and Sem(sample), a luminescence spectral areal intensity when excited at the same wavelength.

[0045] Phosphorescence yield values described herein are relative values with respect to a phosphorescence yield $\delta = 1$ of $lr(ppy)_3$ as a standard sample.

[0046] Further, the luminescence (phosphorescence) life referred to herein is based on values measured according to the following method.

<<Method of measurement of life>>

[0047] A sample compound is dissolved in chloroform and spin-coated onto a quartz substrate in a thickness of ca. 0.1 μ m and is exposed to pulsative nitrogen laser light at an excitation wavelength of 337 nm at room temperature by using a luminescence life meter (made by Hamamatsu Photonics K.K.). After completion of the excitation-pulses, the decay characteristic of luminescence intensity is measured.

[0048] When an initial luminescence intensity is denoted by I₀, a luminescence intensity after t(sec) is expressed according to the following formula with reference to a luminescence life τ(sec):

$$I = I_0 \cdot \exp(-t/\tau).$$

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[0049] Thus, the luminescence life τ is a time period in which the luminescence intensity I is attenuated down to 1/e of the initial intensity I ($II_0 = e^{-1}$, e is a base of natural logarithm).

[0050] A short phosphorescence life is a condition for providing an EL device of a high luminescence efficiency. More specifically, a long phosphorescence life means abundant presence of molecules in a triplet excited state waiting for the luminescence leading to a problem of a lowering in luminescence efficiency particularly at a high current density. The material of the present invention is a suitable luminescence material for an EL device because of a high phosphorescence yield and a short phosphorescence life. Further, it is assumed that because of a short phosphorescence life, the duration at a triplet state is shortened to suppress the concentration extinction. A high stability of the luminescence material of the present invention was also exhibited in an actual current conduction test of actual devices.

[0051] In the case of a phosphorescent material, the luminescence characteristic thereof is severely affected by its molecular environment. In the case of a fluorescence device, the basic property of a luminescence material is examined based on photoluminescence. In the case of phosphorescence, however, the photoluminescence performance does not directly lead to the luminescence performance of an EL device since it is frequently affected by the polarity of host molecules, temperature and solid/liquid state. As a result, EL device performances except for a part thereof cannot be estimated from the photoluminescence result.

[0052] In the case of a ligand of the present invention having a cyclic group having one or plural fluorine atoms, it becomes possible to shift the luminescence wavelength to a shorter side or a longer side because of a change in energy gap. If it is assumed for convenience that HOMO/LUMO of metal electron orbits and HOMO/LUMO of ligand electron orbits can be considered separately, it is understood that MOHO/LUMO energy levels of ligand electron orbits are changed by fluorine atoms having a large electronegativity to change the energy gap between the HOMO level of the metal and the LUMO level of the ligand, thereby shifting the luminescence from the MCLT state as the lowest excited state to a shorter wavelength side or a longer wavelength side. Accordingly, while a luminescence material exhibiting a stably high quantum yield over a broad wavelength range (blue to red) has not been found, it can be realized by a luminescence material of the present invention, thus being able to provide a luminescence material which shows a high efficiency at a desired emission wavelength over a board wavelength range (from blue to red).

[0053] When a device is formed, due to a large electronegativity of fluorine atoms, the inter-molecular interaction is suppressed to physically result in a suppressed crystallinity favoring a uniform film formation and physically suppressing the dimerization reaction to prohibit the energy deactivation leading to an improved luminescence efficiency, thus resulting in an improved electrical property and an improved device stability.

[0054] Further, in the case of using a ligand containing a plurality of fluorine atoms or polyfluoroalkyl groups as substituents, it is considered that direct interaction between luminescence molecules is suppressed due to electrical repulsion caused by their electrical effect or due to steric hindrance, thereby preventing energy deactivation and con-

centration extinction.

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[0055] Further, from the viewpoint of device preparation, a luminescence material having a substituent, particularly a fluorinated substituent, allows easier vacuum deposition due to a lowering in sublimation temperature in the film formation by vacuum deposition, thus providing a great advantage also in this respect.

[0056] As a result, as shown in Examples described hereinafter, by using a luminescence material having a substituent according to the present invention, a stable luminescence for long hours with suppressed concentration extinction can be expected. Further, it becomes possible to attain a high phosphorescence yield over a temperature range of from -20 °C to 60 °C as an actual operation temperature range of an organic luminescence device. Further, in the case of using a compound at a concentration of 8 wt. % or higher with respect to a host material in the luminescence layer or at a concentration higher than a compound having no substituent, it becomes possible to provide an EL device exhibiting excellent luminescence performance while suppressing the concentration extinction. The concentration of the luminescence material of the present invention in the luminescence layer may be at least 8 wt. %, preferably 10 wt. % or higher, but the luminescence material potentially has a possibility of being used even at 100 % without-causing substantial concentration extinction.

[0057] Herein, the term "luminescence performance" refers to a characteristic attributable to a maximum luminescence efficiency which can be expressed as any of a maximum luminance, a maximum of luminance/current, a maximum of light flux/power consumption or a maximum of external quantum yield.

[0058] A high-efficiency luminescence device according to the present invention is applicable to a product requiring energy economization or a high luminance. More specifically, the luminescence device is applicable to a display apparatus, an illumination apparatus, a printer light source or a backlight for a luminescence layer display apparatus. As for a display apparatus, it allows a flat panel display which is light in weight and provides a highly recognizable display at a low energy consumption. As a printer light source, the luminescence device of the present invention can be used instead of a laser light source of a laser beam printer. Independently addressable devices are arranged in an array form to effect a desired exposure on a photosensitive drum thereby forming an image. The apparatus volume can be remarkably reduced by using the devices of the present invention. For the illumination apparatus or backlight, the energy economization effect according to the present invention can be expected.

[0059] For the application to a display, a drive system using a thin-film transistor (abbreviated as TFT) drive circuit according to an active matrix-scheme, may be used. By driving a display panel using a luminescence material of the present invention in a luminescence layer, it becomes possible to allow a stable display for long hours at a good picture quality.

[0060] Hereinbelow, some specific structural formulae of metal coordination compounds represented by the formula (1) used in the present invention are shown in Table 1 appearing hereinafter, which are however only representative examples and are not exhaustive. Ph - P9 used in Table 1 represent partial structured shown below, of which substituents R1, R2, ... ate shown as A-R1, A-R2, ... when contained in the cyclic group A in the formula (1) and as 8-R1, B-R2, ... when contained in the cyclic group B, ... in Table 1.

Ph:
$$R_1$$
 R_2 R_1 R_2 R_1 R_2 R_1 R_2 R_1 R_2 R_2 R_3 R_2 R_4 R_4 R_5 R_4 R_5 R_4

Np1:
$$R_4$$
 R_2 R_4 R_3 R_2 R_4 R_4 R_5 R_2 R_4 R_5 R_6 R_6

FI: K2 Cn1: R_1 Cn2: R_1 Fn1: R_1 R_2 R_1

Table 1-1

													_	
	No	M	m	n (_ ^]	В	A-RI	A-R2	A-R3	A-R4	B-R5	B-R6	8-R7	8-R8
5	1	V 1	3	0	Ph	PI	СНЭ	н	H	н	н	HI	H	H
			- 1	0	Ph	PI		CH3	- ;;	H	- 11 	$\frac{n}{H}$	H	H
	3	<u> </u>	3	- 	Ph	Fi	- H- 	H H	cks	유	유닉	ㅠ	유	# 1
	171	-	3	- 6 - 1	Ph	PI	유	- 	H	cia l	H +	- 1	유	#
	1 3 1	· F	-	ö	Ph	PI	CZHS	- 	H	H .	- 	ㅠ	유	
														H
	6	4	3	0	Ph	<u>P1</u>	H	C2H5	H	H	<u>H</u>	<u>#</u>	H	<u>H</u>
10	7	- 1		0	Ph	P1	- 11	н	C2H5	H	H	H .	H	H
	6	- 5-	3	-	Ph	1	H	H	<u>H</u>	C2H5	변	H	H	H
	9	-	-3	0	Ph	P1	C3H7	H C3H7	<u>H</u>	H	H	H	H .	H
	10	F	3	- 6	Ph Ph	PI	H	H I	H C3H7	- 	H	- 	H	#
	11	-	-3	*	Ph	Pi	H	H 1		C3H7	규	뀨	- 	#
	12	b	3	-	Ph	PI	C4H9	- 1	H	H I	- # -	뀨	뀨	뀨
	13		3	-	Ph	PI		C4H9	- ii 	유	. n	- 11 	H	-
15	14	<u> </u>	-3	-	Ph	PI	H H	H	C4H9	유	H	규	H	H
	15		-3-1	- 6	Ph	Pi	유	- 1		C4H9	유	- H 	H	H
	16	<u> </u>	-	- 6	Ph	Pi	C8H13	 	H H	H	H	뀨	Я	- 11
	17		-;-	-	Ph	PI	Conta	C6H13	 	규	- 	- ii -	Ĥ	#
	19		-;-	-	Ph	Pi	H	H	CSHI3	- #	 	규니	- 17	유
	20	- 5	-3-1	-	Ph	PI	- # -	H	H	C6H13	- 2 	H.	<u> </u>	유
	21	- 6	-	_ 0	Ph	PI	C8H17	- 4		H	H	H	H	H
20	22	- ir	- 3	-	Ph	Pi	H	C8H17.	 2	Н Н	 	- H	H	H
	23	<u> </u>	3	ö	Ph	PI	H	H	C8H17	Н	H	H	H	H
	24	1	3	ō	Ph	PI	H	H	Н	C8H17	Ĥ	H	H	H
	25	F	3	ò	Ph	PI	C12H25	H	H	н	H	H	K	H
	26	ŀ	3	0	Ph	PI	н	C12H25	Н	н	Н	H	H	#
	27	b	3	0	Ph	PI	H	н	C12H25	Н	H	Н	Н	Н
	28	4	3	0	Ph	Pl	н	Н	24	C12H25	Н	H	Ŧ	H
25	29	4	3	0	Ph	PI	C15H31	H	H	H	Н	Н	H	Н
•	30	lr .	3	0	£	P	H	C15H31	H	2	H	H	H	¥
	31	j.	3	0	Ph	PI	Н	Н	C15H31	z	H	Н	Н	H
	32	ŀ	3	0	Ph	PI	H	H	CH3O	H	Н	¥	H	H
	33	-	3	0	Ph	힏	Н	Н	C2H5O `	¥	H	H	Н	Н
	34	lr .	3	0	Ph	PI	H	Ξ_	C4H9O	×	Н	Ξ	H	H
	35	3r	3	9	_Ph	P1	H	C4H9O	H	H	H	×	H	Щ
30	30	1	3	0	Ph	PI	1 н 1		н	н	н	н	Н	1 # 1
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	37	F	3	0	Ph	PI	н	α	н	H] н	H	Н	H
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35	34	•	•		" "	PI	1 "	н) "	, "	, "	"	"
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	40	b	3	_ 0	Ph	PL	H	Н	CF30	H	H	H	H	1 H
	41	b	3	0	Ph	PI	H	Н	C4F9	H	H	Н	H	T T
	42		3	0	Ph.	PI	Н.	C4F9	H	H	T.H.	H		
40	43	1	3	0	Ph	PI	H	<u>H</u>	CZF5CH2O	 !!-	1-#-	H.	H	1 1
	44	b	3	0	Ph Ph	PI	H	C2F5	H	1	#	H	H	1-3-
	48	1	3	- ŏ -	Ph	PI	H	<u> </u>	C2F5 C3F11	H	1-2-	1 11	1_H	* * * * *
	47	1	3	ð	Ph	Pi	H	H	C8F17	1 H	- #	1 #	H	1 H
	48	1	3	0	Ph	PI	H	н	C2F5C2H4	H	Н	H	H	H
	49	1	_3	0	Ph	PI	CHD	Н	CHS	H	H	H	H	H
	50	1	3	_0	_Ph	PI	H.	CHS	CH3	H	H	T H	I H	H
45	51	F	3	0	Ph	PI	C2H5	_H	C2H5	H	Н	Н	H	H
	52	1	3	0	Ph	PI	C4H9	H	C4H9	H	H	Н	H	" H
	53		3	0	Ph	PI	H	C4H9	H	H	H	H	H	H
	54	1	3	0	Ph	PI	H	H	H	H	CAB		H	H
	85	-	3	0	Ph	PI	H	H	H	<u> </u>	H	CX3		Н
	56	1	13	<u> </u>	Ph.	PI	H	Н	<u>H</u>	1 #	1 H	1 #	CHB	
_	57	1	3	0	Ph	PI	H	H	H	H	H	H H	Η	CHB
50	58	+	1-3-	8	Ph.	P1	 H	<u></u>	H	H H	CAH	CH	H H	 변
	59	1 -	3	8	Ph.		H	H	H	H	1 #	H	CAH	H
	_ 0 0				Ph	PI				1 7			100	DI H

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10		B-R8	C4H9	I	I	I	C8H17	I	Ι	I	CH3	r	I	I	CH3	I	н	I	СНЗ	н	I	H	H.	I	Ι	СНЗ	I
. 15		B-R7	н	I	I	CBH17	I	I	I	CH3	I	I	I	CH3	I	I	I	СНЗ	I	СНЗ	Ŧ	СНЗ	H	I.	СНЗ	I	I
		B-R6	Н	н	C8H17	I	I	н	CH3	I	I	I	왕	I	н	н	СНЗ	I	I	I,	CH3	н	I	СНЗ	Н	H	Ι
20		B-R5	н	C8H17	ı	I	I	СНЗ	I	I	I	CH3	I	I.	I	СНЗ	I	I	н	I	H	I.	СНЗ	I	H	I,	СНЗ
25		A-R4	Н	I	I	I	н	н	I	I	r	I	I	I	I	I	I	H	I,	СНЗ	H	I.	I	H	I	I	I
	1-2	A-R3	Н	I	r	I	I	I	r	I	I	I	r	I	I	СНЗ	CH3	СНЗ	СНЗ	I	I	I	I	I	I	Ŧ	C2H5
30	Table 1-2	A-R2	Н	Ŧ	I	I	I	Ι	I	H	I,	CH3	СНЗ	СНЗ	СНЗ	I	Ι	н	Ι	I	I	I	C2H5	C2H5	C2H5	CZHS	x
35		A-R1	H	н	н	I	Ι	СНЗ	СНЗ	ČH3	СНЗ	I	I	I	I	I	I	н	H	I,	C2H5	C2H5	Ι.	I	1	I	I
		В	P1	P1	P1	P1	P1	P1	P1	P1	P1	P1	<u>P</u>	PI	Ы	P1	٤	P1	P1	P1	Ā.	Ы	P	P.	P1	P.	۳.
40		A	Ph	Ph	Ph	Ph	Ph	Ъ	Ph	Ρh	Ph	Ph	Ph	Ph	Ph	Ph	F.	Ph	Ph	-Ph	F.	F.	문	됩	Ph	£	Æ
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10		B-R8	н	н	CH3	I	I	I	I	I	I	CH3	I	I	ı	I	H	н	I	I	СНЗ	I	I	I	Ŧ	I	I
15		B-R7	I	СНЗ	I	İ	СНЗ	I	I	I	CH3	I	I	CH3	I	I	н	I	π.	СНЗ	I	Ι	CH3	1	ı	I	I
••		B-R6	СНЗ	I	I	당	I	왕	Ι	왕	I	Ι,	CH3	ı.	СНЗ	СНЗ	СНЗ	Η,	СНЗ	I	I	СНЗ	I	СНЗ	СНЗ	I	I
20	:	B-R5	I	I	I	I	I	I,	CH3	I	I	I	I	1	I	I	I	СНЗ	H.	Η	H	H	I	I	I	CF3	CF3
25		A-R4	Т	Ι	r	C2H5	C2H5	I	Ι	I	I	I	I	I	C4H9	C4H9	I	I	I	I	I	Ι	н	С6Н13	СБН13	н	1
	(continued)	A-R3	C2H5	C2H5	C2H5	I	I	I	I	I	I	Ι.	C4H9	C4H9	I	I	I	I	I	I	I	С6Н13	C6H13	Ι	I	I	I
30	Table 1-2 (A-R2	I	I	I	I	I	I	C4H9	C4H9	C4H9	C4H9	I	I	I	I	I	Сен13	СБН13	C6H13	С6Н13	1	I	Ι	I	I	CH3
35	12	A-R1	I	I	I	Ή	I.	C4H9	I	I	I.	I	I	I	Ι	I	Сен13	I	I	I	I	I	I	I	Ι	СНЗ	I
		В	Ы	P1	P1	P1	P1	P1	P1	P1	P1	P1	P.	٦.	P1	2	2	F	E	٦	ā	<u>.</u>	2	٦	7	7	E
40		⋖	Ph H	Ph	Ph	Ph	Ph	Ph	Ph	Ph	Ph	Ph	H.	Æ	ብ:	듄	£	듄	£	뚭	£	H.	F.	£	Æ	듄	듄
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50		å	98	87	88	68	66	91	92	93	94	98	96	26	86	8	5	101	102	103	104	105	106	107	108	109	110

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10		B-R8	I	I	Ι	I	I	I	I	I	I	I
15		B-R7	I	r	I	I	I	I	CF3	CF3	CF3	CF3
		B-R6	н	н	CF3	CF3	CF3	CF3	I	I	I	Ι
20		B-R5	CF3	CF3	I	I	I	I	I	I	I	I
25	-	A-R4	т	CH3	I	Ξ	Ξ	CH3	I	I	I	왕
	continued)	A-R3	СНЗ	I	I	I	CH3	I	Ι	I	CH3	I
30	Table 1-2 (continued)	A-R2	I	I	I	CH3	Ξ	I	I	CH3	I	I
35	Ta	A-R1	Ξ	I	CH3	I	I	I	CH3	I	I	I
		В	<u>P</u>	P4	Ē	٦	7	P	7	٩	٣	2
40		٧	H.	H.	뚭	돈	윤	F.	문	P.	P.	듄
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Table 1-3

							,	Table 1-						
	No	М	m	п	Α	В	A-R1	A-R2	A-R3	A-R4	B-R5	B-R6	B-R7	B-R8
5	121	lr	3	0	Ph	P1	СНЗ	Н	Н	H	Н	Н	Н	CF3
	122	lr	3	0	Ph	P1	Н	СНЗ	н	Н	Н	Н	н	CF3
	123	lf	3	0	Ph	P1	Н	Н	СНЗ	Н	H	Н	Н	CF3
	124	lr	3	0	Ph	P1	Н	Н	Н	СНЗ	Н	Н	Н	CF3
10	125	lr	3	0	Ph	P1	СНЗ	Н	н	Н	F	н	н	Н
	126	lr	3	0	Ph	P1	н	СНЗ	н	н	F	Н	Н	Н
	127	lr	3	0	Ph	P1	Н	Н	СНЗ	Ħ	F	Ŧ	Н	Н
15	128	lr	3	0	Ph	P1	Н	н	Н	СНЗ	F	Н	Н	Н
	129	r	3	0	Ph	P1	СНЗ	Н	н	Η	H	F	Н	Н
	130	ŀr	3	0	Ph	P1	Н	СНЗ	Н	Ι	Н	F	I	H
	131	lr	3	0	Ph	P1	Н	Н	СНЗ	Η	Τ	Ŧ	Н	Н
20	132	lr	3	0	Ph	P1	Н	Н	Н	CH3	H	F	н	Н
	133	lr	3	0	Ph	P1	СНЗ	Н	Н	Ή	Ι	н	F	Н
	134	lr	3	0	Ph	P1	Н	СНЗ	Н	Н	Н	Н	F	Н
25	135	lr	3	0	Ph	P1	Н	Н	СНЗ	Н	Н	Н	F	Η .
	136	lr	3	0	Ph	P1	Н	Н	Н	СНЗ	Н	Н	F	Н
	137	lr	3	0	Ph	P1	СНЗ	Н	Н	Н	Н	Н	Н	F
	138	lr	3	0	Ph	P1	Н	СНЗ	н	Н	Н	Н	Н	F
30	139	ir	3	0	Ph	P1	Н	н	СНЗ	Н	Н	Н	Н	.F
	140	lr	3	0	Ph	P1	Н	Н	Н	СНЗ	Н	Н	Н	F
	141	lr	3	0	Ph	P1	C2H5	Н	Н	Н	CF3	Н	Н	Н
35	142	lr	3	0	Ph	P1	Н	C2H5	Н	Н	-CF3	Н	Н	Н
	143	lr	3	0	Ph	P1	Н	Н	C2H5	Н	CF3	Н	Н	Н,
	144	lr	3	0	Ph	P1	Н	Н	Н	C2H5	CF3	Н	Н	н
40	145	lr	3	0	Ph	P1	C2H5	Н	н	Н	Н	CF3	Н,	Н
40	146	lr	3	0	Ph	P1	Н	C2H5	н	Н	Н	-CF3	Н	Н
	147	lr	3	0	Ph	P1	Н	H	·C2H5	Н	Н	·CF3	Н	Н
	148	lr	3	0	Ph	P1	Н	Н	н	C2H5	Н	-CF3	Н	Н
45	149	lr .	3	0	Ph	P1	C2H5	Н	Н	Н	Н	н	°CF3	н
	150	lr	3	0	Ph	P1	Н	C2H5	Н	Н	Н	Н	-CF3	Н
	151	lr	3	0	Ph	P1	Н	Н	C2H5	Н	Н	н	·CF3	Н
50	152	lr	3	0	Ph	P1	Н	Н	н	·C2H5	Н	н	·CF3	Н
~	153	lr	3	0	Ph	P1	C2H5	Н	Н	Н	Н	Н	Н	·CF3
	154	lr	3	0	Ph	P1	H	C2H5	Н	Н	Н	Н	Н	CF3
	155	lr	3	0	Ph	P1	Н	Н	C2HS	Н	Н	н	Н	'CF3
55	156	lr	3	0	Ph	P1	H	Н	Н	C2H5	Н	н	Н	CF3
	157	lr	3	0	Ph	P1	C2H5	I	Н	Н	F	Н	Н	Н
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Table 1-3 (continued)

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	No	М	m	n	Α	В	A-R1	A-R2	A-R3	A-R4	B-R5	B-R6	B-R7	8-R8
	158	lr	3	0	Ph	P1	Н	C2H5	Н	Н	F	Н	Н	Н
5	159	lr	3	0	Ph	P1	Н	н	C2H5	H	F	Н	Н	Н
	160	lr	3	0	Ph	P1	Н	Н	Н	C2H5	Ħ,	Н	Н	Н
	161	lr	3	0	Ph	P1	C2H5	Н	Н	Н	Н	F Î	Н	Н
10	162	lr	3	0	Ph	P1	н	C2H5	Н	Н	Ħ	F	Н	Н
	163	lr	3	0	Ph	P1	Н	Н	C2H5	Н	Н	F	Н	Н
	164	lr	3	0	Ph	P1	Н	Н	H	C2H5	Н	F	Н	Н
	165	lr	3	0	Ph	P1	C2H5	Н	Н	Н	Н	Н	F	Н
15	166	lr	3	0	Ph	P1	Н	C2H5	Н	Н	Н	Н	F	Н
	187	lr	3	0	Ph	P1	Н	Н	C2H5	Н	Н	Н	F	Н
	168	lr	3	0	Ph	P1	Н	Н	Н	C2H5	Н "	Н	F	Н
20	169	lr	3	0	Ph	P1	C2H5	Н	Н	н	H	Н	Н	F
	170	lr	3	0	Ph	Pi	Н	C2H5	Н	Н	Н	Н	Н	F
	171	lr	3	0	Ph	P1	Н	Н	C2H5	Н	Н	Н	Н	F
	172	lr	3	0	Ph	P1	Н	Н	Н	C2H5	H	н	Н	F
25	173	lr	3	0	Ph	P1	C4H9	Н	Н	Н	F	Н	Н	Н
	174	lr	3	0	Ph	P1	Н	C4H9	Н	H	F	Н	Н	H Î
	175	lr	3	0	Ph	P1	Н	Н	C4H9	Н	F	Н	Н	Н
30	176	lr.	3	0	Ph	P1	Н	Н	Н	C4H9	F	Н	H,	Н
	177	lr	3	0	Ph	P1	C4H9	Н	Н	Н	н	F	Н	Н
	178	ir	3	0	Ph	P1	Н	C4H9	Н	Н	Н	F	Н	Н
25	179	lr	3	0	Ph	P1	Н	Н	C4H9	Н	н	F	Н	Н
35	180	lr	3	0	Ph	P1	Н	Н	н	C4H8	н	F	Н	H

Table 1-4

								lable 1	-4					
40	No	М	m	n	Α	В	A-R1	A-R2	A-R3	A-R4	B-R5	B-R6	B-R7	B-R8
	181	lr	3	0	Ph	P1	C4H9	Н	н	Н	Н	Н	F	Н
	182	lr	3	0	Ph	P1	Н	C4H9	Н	Н	H	Н	F	Н
45	183	lr	3	0	Ph	P1	н	н	C4H9	Н	H	Ξ	F	Н
	184	lr	3	0	Ph	P1	Н	н	Н	C4H9	Н	Н	F	Н
	185	Ir	3	0	Ph	P1	C4H9	Н	Н	Н	Н	Н	Н	F
	186	lr	3	0	Ph	P1	н	C4H9	Н	Н	H	Н	H	F
50	187	lr	3	0	Ph	P1	н	Н	C4H9	Н	Н	Н	H	F
	188	lr	3	0	Ph	P1	н	Н	Н	C4H9	Н	Н	Н	F
	189	Ir	3	0	Ph	P1 .	C4H9	н	н	Н	CF3	Н	Н	Н
<i>55</i>	190	lr	3	0	Ph	P1	Н	C4H9	Н	Н	CF3	Н	Н	Н
	191	lr	3	0	Ph	P1	Н	Н	C4H9	н	CF3	н	Н	Н
	192	lr	3	0	Ph	P1	Н	Н	Н	C4H9	CF3	Н	н	Н

Table 1-4 (continued)

193 If 3 0 Ph P1 C4H9 H H H CF3 H T CF3 H							,		10.14 (0.	1 3		· · · · ·			
194 Ir 3 0 Ph P1 H C4H9 H H CF3 H H H H F H H H H H						-		A-R1						B-R7	B-R8
194 IF 3 0 0 Ph P1	_	193	lr .	3	0	Ph	P1	C4H9	Н	Н	н	Н	CF3	Н]	Н
196 Ir 3 0 Ph P1 H H H C4H9 H CF3 H H 197 Ir 3 0 Ph P1 C4H9 H H H H H CF3 H H 198 Ir 3 0 Ph P1 H C4H9 H H H H CF3 H H 199 Ir 3 0 Ph P1 H H H C4H9 H H H CF3 H C20 Ir 3 0 Ph P1 H H H H C4H9 H H H CF3 H C20 Ir 3 0 Ph P1 H H H H H H H H CF3 H C20 Ir 3 0 Ph P1 H C4H9 H H H H H CF3 H C20 Ir 3 0 Ph P1 H C4H9 H H H H H CF3 H C20 Ir 3 0 Ph P1 H H C4H9 H H H H H CF3 H C5 C20 Ir 3 0 Ph P1 H H H C4H9 H H H H H C5 C20 Ir 3 0 Ph P1 H H H C4H9 H H H H C5 C20 Ir 3 0 Ph P1 H C8H17 H H F H H H C5 C20 Ir 3 0 Ph P1 H C8H17 H H F H H H H H H H	5	194	lr	3	0	Ph	P1	Н	C4H9	н	Н	Н	-CF3	_ н]	н
197 Ir 3 0 Ph P1 C4H9 H H H H H CF3 H 198 Ir 3 0 Ph P1 H C4H9 H H H H CF3 H 199 Ir 3 0 Ph P1 H C4H9 H H H H CF3 H 200 Ir 3 0 Ph P1 H H H C4H9 H H H CF3 H 201 Ir 3 0 Ph P1 H H H H H H H CF3 H 202 Ir 3 0 Ph P1 C4H9 H H H H H H CF3 H 203 Ir 3 0 Ph P1 H C4H9 H H H H H CF3 204 Ir 3 0 Ph P1 H C4H9 H H H H H CF3 205 Ir 3 0 Ph P1 H H C4H9 H H H H CF4 205 Ir 3 0 Ph P1 H C8H17 H H F H H H 206 Ir 3 0 Ph P1 H C8H17 H H F H H H 207 Ir 3 0 Ph P1 H C8H17 H H F H H H 209 Ir 3 0 Ph P1 H C8H17 H H H F H H 209 Ir 3 0 Ph P1 H C8H17 H H H F H H 210 Ir 3 0 Ph P1 H C8H17 H H H F H H 211 Ir 3 0 Ph P1 H C8H17 H H H F H H 212 Ir 3 0 Ph P1 H C8H17 H H H F H H 213 Ir 3 0 Ph P1 H C8H17 H H H H F H 214 Ir 3 0 Ph P1 H C8H17 H H H H H F H 215 Ir 3 0 Ph P1 H C8H17 H H H H H F H 216 Ir 3 0 Ph P1 H C8H17 H H H H H F H 217 Ir 3 0 Ph P1 H C8H17 H H H H H H F H 218 Ir 3 0 Ph P1 H C8H17 H H H H H H F H 219 Ir 3 0 Ph P1 H C8H17 H H H H H H H F 220 Ir 3 0 Ph P1 H C8H17 H H H H H H H F 221 Ir 3 0 Ph P1 H C8H17 H H H H H H H F 222 Ir 3 0 Ph P1 C8H17 H H H C8H17 H H H H H H H 223 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H H 224 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H H 225 Ir 3 0 Ph P1 H C8H17 H H H CF3		195	lr	3	0	Ph	P1	Н	н	C4H9	Н	н	CF3	Н	н
198 ir 3 0 Ph P1 H C4H9 H H H H CF3 H 199 ir 3 0 Ph P1 H H C4H9 H H H H CF3 H 199 ir 3 0 Ph P1 H H H C4H9 H H H C53 H 200 ir 3 0 Ph P1 H H H H C4H9 H H H C53 H 201 ir 3 0 Ph P1 C4H9 H H H H C53 H 201 ir 3 0 Ph P1 H H H H H H H H H C53 H 202 ir 3 0 Ph P1 H C4H9 H H H H H H C53 203 ir 3 0 Ph P1 H H H C4H9 H H H H H H C53 204 ir 3 0 Ph P1 H H H C4H9 H H H H H C53 205 ir 3 0 Ph P1 C8H17 H H H F H H H C53 206 ir 3 0 Ph P1 H C8H17 H H F H H H H H 206 ir 3 0 Ph P1 H C8H17 H H F H H H H 207 ir 3 0 Ph P1 H H H C8H17 H F H H H 208 ir 3 0 Ph P1 H H H C8H17 H H F H H H 209 ir 3 0 Ph P1 H C8H17 H H H F H H H H 210 ir 3 0 Ph P1 H C8H17 H H H F H H H 210 ir 3 0 Ph P1 H C8H17 H H H H F H H 211 ir 3 0 Ph P1 H C8H17 H H H H F H H 212 ir 3 0 Ph P1 H C8H17 H H H H F H H 213 ir 3 0 Ph P1 H C8H17 H H H H F H H 214 ir 3 0 Ph P1 H C8H17 H H H H F H H 215 ir 3 0 Ph P1 H C8H17 H H H H H F H H 216 ir 3 0 Ph P1 H C8H17 H H H H H F H H 217 ir 3 0 Ph P1 H C8H17 H H H H H H F H 218 ir 3 0 Ph P1 H C8H17 H H H H H H F H 219 ir 3 0 Ph P1 H C8H17 H H H H H H F H 210 ir 3 0 Ph P1 H C8H17 H H H H H H H F H 211 ir 3 0 Ph P1 H C8H17 H H H H H H H F H 212 ir 3 0 Ph P1 H C8H17 H H H H H H H F H 213 ir 3 0 Ph P1 H C8H17 H H H H H H H F H 214 ir 3 0 Ph P1 H H C8H17 H H H H H H H F H 215 ir 3 0 Ph P1 H H H C8H17 H H H H H H H F H 216 ir 3 0 Ph P1 H H C8H17 H H H H H H H H F 217 ir 3 0 Ph P1 H H H H C8H17 H H H H H H H H H H H 228 ir 3 0 Ph P1 H H H H C8H17 H H H H H H H H H H H 229 ir 3 0 Ph P1 H H H H C8H17 H H H H H H H H H H H 220 ir 3 0 Ph P1 H H H H C8H17 H H H H H H H H H H H 220 ir 3 0 Ph P1 H H H H C8H17 H H H H H H H H H H H 220 ir 3 0 Ph P1 H H H H C8H17 H H H H H H H H H H H 220 ir 3 0 Ph P1 H H H H H H H H H H H H H H H H 220 ir 3 0 Ph P1 H H H H H H H H H H H H H H H H H H		196	lr	3	0	Ph	P1	н	н	Н	C4H9	H	CF3	Н	Н
199 ir 3 0 Ph P1 H H C4H9 H H H CF3 H 200 ir 3 0 Ph P1 H H H H C4H9 H H CF3 H 201 ir 3 0 Ph P1 H H H H H H H H CF3 H 202 ir 3 0 Ph P1 C4H9 H H H H H H CF3 H 203 ir 3 0 Ph P1 H C4H9 H H H H H CF3 204 ir 3 0 Ph P1 H H C4H9 H H H H CF3 205 ir 3 0 Ph P1 C8H17 H H H F H H H 206 ir 3 0 Ph P1 H C8H17 H H F H H H 207 ir 3 0 Ph P1 H H C8H17 H F H H H 208 ir 3 0 Ph P1 H H H C8H17 F H H H 209 ir 3 0 Ph P1 C8H17 H H H H F H H 210 ir 3 0 Ph P1 H C8H17 H H H F H H 211 ir 3 0 Ph P1 H C8H17 H H H F H H 212 ir 3 0 Ph P1 H H H C8H17 H H F H H 213 ir 3 0 Ph P1 H H H C8H17 H H H F H 214 ir 3 0 Ph P1 H H C8H17 H H H F H 215 ir 3 0 Ph P1 H H C8H17 H H H H F H 216 ir 3 0 Ph P1 H H H C8H17 H H H H F H 217 ir 3 0 Ph P1 H H H C8H17 H H H H H F H 218 ir 3 0 Ph P1 H H H H H H H H H	10	197	lr	3	0	Ph	P1	C4H9	Н	н	H	Н	H	-CF3	Н
200 Ir 3 0 Ph P1 H H H H C4H9 H H CF3 H		198	lr	3	0	Ph	P1	Н	C4H9	Н	Н	Н	H	CF3	Н
201 ir 3 0 Ph P1 C4H9 H H H H H H H CF		199	lr	3	0	Ph	P1	Н	н	-C4H9	Н	H	Н	CF3	Н
201 3 0 Ph P1 CAH9 H H H H H H CF CAH9 H H H H H H CF CF CAH9 H H H H H CF CF CAH9 H H H H H CF CF CAH9 CF CAH9 H H H H CF CF CAH9 H H H CF CF CAH9 H H H CF CF CAH9 H H H H CF CF CAH9 H H H H CF CF CAH9 H H H H CF CF CF CAH9 H H H H CF H H H H CF H H H H F H H H H		200	lr	3	0	Ph	P1	Н	Н	н	·C4H9	Н	Н	CF3	Н
203 Ir 3 0 Ph P1 H H C4H9 H H H H CF 204 Ir 3 0 Ph P1 H H H C4H9 H H H CF 205 Ir 3 0 Ph P1 C8H17 H H H F H H H H 206 Ir 3 0 Ph P1 H C8H17 H H F H H H H 207 Ir 3 0 Ph P1 H H C8H17 H F H H H H 208 Ir 3 0 Ph P1 H H H C8H17 F H H H 209 Ir 3 0 Ph P1 C8H17 H H H H F H H 210 Ir 3 0 Ph P1 H C8H17 H H H F H H 211 Ir 3 0 Ph P1 H H C8H17 H H F H H 212 Ir 3 0 Ph P1 H H H C8H17 H H F H H 213 Ir 3 0 Ph P1 C8H17 H H H H H F H 214 Ir 3 0 Ph P1 H C8H17 H H H H F H 215 Ir 3 0 Ph P1 H C8H17 H H H H F H 216 Ir 3 0 Ph P1 H H C8H17 H H H H F H 217 Ir 3 0 Ph P1 H H C8H17 H H H H H F H 218 Ir 3 0 Ph P1 H C8H17 H H H H H H F 219 Ir 3 0 Ph P1 H C8H17 H H H H H H H 220 Ir 3 0 Ph P1 H C8H17 H H H H H H H 221 Ir 3 0 Ph P1 H C8H17 H H H H H H H 222 Ir 3 0 Ph P1 H C8H17 H H H H H H H 223 Ir 3 0 Ph P1 H C8H17 H H CF3 H H H 224 Ir 3 0 Ph P1 H H C8H17 H H CF3 H H 225 Ir 3 0 Ph P1 H H C8H17 H H CF3 H H 226 Ir 3 0 Ph P1 H H H C8H17 H H CF3 H H 226 Ir 3 0 Ph P1 C8H17 H H H H CF3 H H 226 Ir 3 0 Ph P1 C8H17 H H H H CF3 H H 226 Ir 3 0 Ph P1 C8H17 H H H H CF3 H H 227 Ir 3 0 Ph P1 H H H H H C8H17 H H H CF3 H 228 Ir 3 0 Ph P1 H H H H H H CF3 H H 239 Ir 3 0 Ph P1 H H H H H H H H H	15	201	lr	3	0	Ph	P1	C4H9	٠Н	н	Н	Н	Н	Н	CF3
20		202	lr	3	0	Ph	P1	Н	·C4H9	Н	н	Н	Н	Н	CF3
205 Ir 3 0 Ph P1 C8H17 H H H F H H H H 206 Ir 3 0 Ph P1 H C8H17 H H F H H H H 207 Ir 3 0 Ph P1 H H C8H17 H F H H H H 208 Ir 3 0 Ph P1 H H H C8H17 F H H H 209 Ir 3 0 Ph P1 C8H17 H H H H F H H 210 Ir 3 0 Ph P1 H C8H17 H H H F H H 211 Ir 3 0 Ph P1 H H C8H17 H H F H H 212 Ir 3 0 Ph P1 H H H C8H17 H F H 213 Ir 3 0 Ph P1 C8H17 H H H H H F H 214 Ir 3 0 Ph P1 H C8H17 H H H H F H 215 Ir 3 0 Ph P1 H C8H17 H H H H F H 216 Ir 3 0 Ph P1 H H C8H17 H H H H F H 217 Ir 3 0 Ph P1 C8H17 H H H H H H H 218 Ir 3 0 Ph P1 H C8H17 H H H H H H 219 Ir 3 0 Ph P1 H C8H17 H H H H H H 220 Ir 3 0 Ph P1 H C8H17 H H H H H H 221 Ir 3 0 Ph P1 H C8H17 H H H H H H 222 Ir 3 0 Ph P1 H C8H17 H H H H H 223 Ir 3 0 Ph P1 H C8H17 H H CF3 H H 224 Ir 3 0 Ph P1 H C8H17 H H CF3 H H 225 Ir 3 0 Ph P1 H H C8H17 H H CF3 H H 226 Ir 3 0 Ph P1 H H C8H17 H H H CF3 H H 226 Ir 3 0 Ph P1 H H H C8H17 CF3 H H 227 Ir 3 0 Ph P1 H H H C8H17 CF3 H H 228 Ir 3 0 Ph P1 H H H C8H17 H H H CF3 H H 229 Ir 3 0 Ph P1 H H H C8H17 H H H CF3 H H 230 Ir 3 0 Ph P1 H H H H C8H17 H H H CF3 H H 240 Ir 3 0 Ph P1 H H H H C8H17 H H H CF3 H H 241 Ir 3 0 Ph P1 H H H H C8H17 H H H CF3 H H H 242 Ir 3 0 Ph P1 H H H H H H C8H17 H H H H H H H		203	lr	3	0	Ph	P1	Н	Н	C4H9	Н	H	H	Н	CF3
206 Ir 3 0 Ph P1 H C8H17 H H F H H H H C8H17 H F H H H H C8H17 H F H H H H C8H17 H H H H F H H H H F H H	20	204	lr	3	0	Ph	P1	Н	Н	н	C4H9	Н	H	H	CF3
207 Ir 3 0 Ph P1 H H C8H17 H F H H H H C8H17 F H H H H H C8H17 F H H H H H C8H17 F H H H H H H H H H		205	lr	3	0	Ph	P1	C8H17	Н	Н	н	F	Н	н	Н
258 Ir 3 0 Ph P1 H H H C8H17 F H H H H H H H H H		206	lr	3	0	Ph	P1	Н	C8H17	Н	н	F	Н	Η.	Н
208		207	lr	3	0	Ph	P1	Н	Н	C8H17	Н	F	H	Н	Н
210 Ir 3 0 Ph P1 H C8H17 H H H F H H H H F H H	25	208	lr	3	0	Ph	P1	Н	Н	Н	C8H17	F	Н	Н	Н
211 Ir 3 0 Ph P1 H H C8H17 H H F H H H C8H17 H F H H H C8H17 H F H H H C8H17 H F H H H H H H H H		209	lr	3	0	Ph	P1	C8H17	Н	Н	Н	Н	F	Н	Н
212 Ir 3 0 Ph P1 H H H C8H17 H F H H 213 Ir 3 0 Ph P1 C8H17 H H H H H F H 214 Ir 3 0 Ph P1 H C8H17 H H H H F H 215 Ir 3 0 Ph P1 H H C8H17 H H H F H 216 Ir 3 0 Ph P1 H H H C8H17 H H H F H 217 Ir 3 0 Ph P1 C8H17 H H H H H H H H 218 Ir 3 0 Ph P1 H C8H17 H H H H H H H 219 Ir 3 0 Ph P1 H H C8H17 H H H H H H 220 Ir 3 0 Ph P1 H H H C8H17 H H H H H 221 Ir 3 0 Ph P1 C8H17 H H H CF3 H H 222 Ir 3 0 Ph P1 H C8H17 H H CF3 H H 223 Ir 3 0 Ph P1 H H C8H17 H CF3 H H 224 Ir 3 0 Ph P1 H H C8H17 CF3 H H 225 Ir 3 0 Ph P1 H H H C8H17 CF3 H H 226 Ir 3 0 Ph P1 C8H17 H H H H CF3 H H 226 Ir 3 0 Ph P1 C8H17 H H H H CF3 H H 227 Ir 3 0 Ph P1 C8H17 H H H H CF3 H H 248 Ir 3 0 Ph P1 C8H17 H H H H CF3 H H 259 C8 C8 C8 C8 C8 C8 C8 C		210	Ir	3	0	Ph .	P1	Н	CBH17	Н	Н	Н	F	Н	Н
213 Ir 3 0 Ph P1 C8H17 H H H H H F H 214 Ir 3 0 Ph P1 H C8H17 H H H H F H 215 Ir 3 0 Ph P1 H H C8H17 H H H F H 216 Ir 3 0 Ph P1 H H H C8H17 H H F H 217 Ir 3 0 Ph P1 C8H17 H H H H H H F 218 Ir 3 0 Ph P1 H C8H17 H H H H H H F 219 Ir 3 0 Ph P1 H H C8H17 H H H H H F 220 Ir 3 0 Ph P1 H H H C8H17 H H H H F 221 Ir 3 0 Ph P1 C8H17 H H H CF3 H H H 222 Ir 3 0 Ph P1 H C8H17 H H CF3 H H H 223 Ir 3 0 Ph P1 H H C8H17 H CF3 H H H 224 Ir 3 0 Ph P1 H H H C8H17 CF3 H H H 225 Ir 3 0 Ph P1 H H H C8H17 CF3 H H H 226 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 226 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 226 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 227 Ir 3 0 Ph P1 H H H H H CF3 H H 238 Ir 3 0 Ph P1 H H H H H CF3 H H 249 Ir 3 0 Ph P1 H H H H H CF3 H H 240 Ir 3 0 Ph P1 H H H H H CF3 H H 240 Ir 3 0 Ph P1 H H H H H CF3 H H 241 Ir 3 0 Ph P1 H H H H H H H H H	30	211	lr	3	0	Ph	P1	Н	Н	C8H17	н	Н	F	Н	Н
214 Ir 3 0 Ph P1 H C8H17 H H H H F H 215 Ir 3 0 Ph P1 H H C8H17 H H H F H 216 Ir 3 0 Ph P1 H H H C8H17 H H H F H 217 Ir 3 0 Ph P1 C8H17 H H H H H H H F 40 218 Ir 3 0 Ph P1 H C8H17 H H H H H H F 219 Ir 3 0 Ph P1 H H C8H17 H H H H H F 220 Ir 3 0 Ph P1 H H H C8H17 H H H H F 41 221 Ir 3 0 Ph P1 C8H17 H H H CF3 H H H 222 Ir 3 0 Ph P1 H C8H17 H H CF3 H H H 223 Ir 3 0 Ph P1 H H C8H17 H CF3 H H H 224 Ir 3 0 Ph P1 H H H C8H17 CF3 H H H 225 Ir 3 0 Ph P1 C8H17 H H H H CF3 H H 226 Ir 3 0 Ph P1 C8H17 H H H H CF3 H H 226 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 226 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 226 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 226 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 227 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 228 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 249 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 260 Ir 3 0 Ph P1 H H H H H CF3 H H 270 Ir 3 0 Ph P1 H H H H H H H 280 Ir 3 0 Ph P1 H H H H H H 290 Ir 3 0 Ph P1 H H H H H 200 Ir 3 0 Ph P1 H H H H H H 201 Ir 3 0 Ph P1 H H H H H H 201 Ir 3 0 Ph P1 H H H H H H 201 Ir 3		212	lr	3	0	Ph	P1	Н	Н	Н	C8H17	Н	F	Н	Н
215 Ir 3 0 Ph P1 H H C8H17 H H H F H 216 Ir 3 0 Ph P1 H H H C8H17 H H F H 217 Ir 3 0 Ph P1 C8H17 H H H H H H H H H		213	lr	3	0	Ph	P1	C8H17	Н	Н	Н	Н	Н	F	Н
215 Ir 3 0 Ph P1 H H C8H17 H H H F H 216 Ir 3 0 Ph P1 H H H C8H17 H H H F H 217 Ir 3 0 Ph P1 C8H17 H H H H H H H H 218 Ir 3 0 Ph P1 H C8H17 H H H H H H 219 Ir 3 0 Ph P1 H H C8H17 H H H H H 220 Ir 3 0 Ph P1 H H H C8H17 H H H H 221 Ir 3 0 Ph P1 C8H17 H H H CF3 H H 222 Ir 3 0 Ph P1 H C8H17 H H CF3 H H 223 Ir 3 0 Ph P1 H H C8H17 H CF3 H H 224 Ir 3 0 Ph P1 H H C8H17 CF3 H H 225 Ir 3 0 Ph P1 C8H17 H H H CF3 H H 226 Ir 3 0 Ph P1 C8H17 H H H CF3 H H 226 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 226 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 226 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 227 C8H17 CF3 H H H 228 C8H17 C8H17 H H H CF3 H H 229 C8H17 C8H17 H H H H CF3 H H 220 C8H17 C8H17 H H H H CF3 H H 221 C8H17 H H H CF3 H H 2222 C8H17 C8H17 H H H CF3 H H 223 C8H17 C8H17 H H H C8H17 H H CF3 H 240 C8H17 C8H17 H H H CF3 H H 241 C8H17 C8H17 H H H CF3 H H 242 C8H17 C8H17 H H H C8H17 H H H CF3 H H 243 C8H17 C8H17 C8H17 H H H C8H17 CF3 H H 244 C8H17 C8H17 H H H H H C8H17 H H H CF3 H H H C8H17 H H H H H C8H17 H H H H C8H17 H H H C8H17 H H H H H H C8H17 H H H C8H17 H H H H H C8H17 H H H H C8H17 H H H H C8H17 H H H	<u> </u>	214	lr	3	0	Ph	P1	н	C8H17	Н	н	Н	Н	F	Н
217	35	215	lr	3	0	Ph	P1	Н	н	C8H17	Н	Н	Н	F	Н
40 218 ir 3 0 Ph P1 H C8H17 H H H H H F E 219 ir 3 0 Ph P1 H H C8H17 H H H H H F E 220 ir 3 0 Ph P1 H H H C8H17 H H H H F E 221 ir 3 0 Ph P1 C8H17 H H H CF3 H H F E 222 ir 3 0 Ph P1 H C8H17 H H CF3 H H F E 223 ir 3 0 Ph P1 H C8H17 H H CF3 H H F E 224 ir 3 0 Ph P1 H H C8H17 H CF3 H H F E 225 ir 3 0 Ph P1 H H C8H17 H CF3 H H F E 226 ir 3 0 Ph P1 C8H17 H H H C8H17 CF3 H H F E 226 ir 3 0 Ph P1 C8H17 H H H H CF3 H F E 226 ir 3 0 Ph P1 C8H17 H H H H CF3 H F E 226 ir 3 0 Ph P1 C8H17 H H H H CF3 H F E 226 ir 3 0 Ph P1 H C8H17 H H H H CF3 H F E 226 ir 3 0 Ph P1 H C8H17 H H H H CF3 H F E 226 ir 3 0 Ph P1 H C8H17 H H H H H CF3 H F E 226 ir 3 0 Ph P1 H C8H17 H H H H H CF3 H F E 226 ir 3 0 Ph P1 H C8H17 H H H H H CF3 H F E 226 ir 3 0 Ph P1 H C8H17 H H H H H CF3 H F E 226 ir 3 0 Ph P1 H C8H17 H H H H H CF3 H F E		216	lr	3	0	Ph	P1	Н	Н	Н	C8H17	Н	Н	F	Н
219 Ir 3 0 Ph P1 H H C8H17 H H H H F 220 Ir 3 0 Ph P1 H H H C8H17 H H H H F 221 Ir 3 0 Ph P1 C8H17 H H H CF3 H H H 222 Ir 3 0 Ph P1 H C8H17 H H CF3 H H H 223 Ir 3 0 Ph P1 H H C8H17 H CF3 H H H 224 Ir 3 0 Ph P1 H H H C8H17 CF3 H H H 225 Ir 3 0 Ph P1 C8H17 H H H H CF3 H H 226 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 226 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 227 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 228 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 240 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H 250 Ir 3 0 Ph P1 H C8H17 H H H H CF3 H H 250 Ir 3 0 Ph P1 H H H H H H H 250 Ir 3 0 Ph P1 H H H H H H H H H		217	ir	3	0	Ph	P1	C8H17	Н	Н	н	Н	Н	Н	F
220 Ir 3 0 Ph P1 H H H C8H17 H H H F 221 Ir 3 0 Ph P1 C8H17 H H H CF3 H H H 222 Ir 3 0 Ph P1 H C8H17 H H CF3 H H H 223 Ir 3 0 Ph P1 H H C8H17 H CF3 H H H 224 Ir 3 0 Ph P1 H H H C8H17 CF3 H H H 225 Ir 3 0 Ph P1 C8H17 H H H H CF3 H H 226 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 226 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 227 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 228 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 229 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 240 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 241 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H 250 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H H 250 Ir 3 0 Ph P1 H H H H C8H17 H H H CF3 H H H 250 Ir 3 0 Ph P1 H H H H H H C8H17 H H H H H H H H H	40	218	lr	3	0	Ph	P1	Н	CBH17	Н	Н	Н	Н	н	F
221 Ir 3 0 Ph P1 C8H17 H H H CF3 H H H F F C22 Ir 3 0 Ph P1 H C8H17 H H CF3 H H H F F C23 Ir 3 0 Ph P1 H H C8H17 H CF3 H H F F C74 F F F F F F F F F		219	lr	3	0	Ph	P1	н	Н	C8H17	н	Н	Н	Н	F
222 3 0 Ph P1		220	lr	3	0	Ph	P1	Н	н	н	·C8H17	н	Н	Н	F
222 Ir 3 0 Ph P1 H C8H17 H H CF3 H H H F1 F2 F2 F2 F2 F2 F2 F2	15	221	lr	3	0	Ph	P1	C8H17	н	н	Н	CF3	н	н	Н
224 Ir 3 0 Ph P1 H H H C8H17 CF3 H H H C25 Ir 3 0 Ph P1 C8H17 H H H H CF3 H CF3 H H CF3	7	222	ir	3	0	Ph	P1	Н	C8H17	Н	Н	CF3	Н	Н	Н
50 225 Ir 3 0 Ph P1 C8H17 H H H CF3 H H 226 Ir 3 0 Ph P1 H C8H17 H H H CF3 H H		223	ŀr	3	0	Ph	P1	Н	Н	C8H17	Н	CF3	Н	Н	Н
226 Ir 3 0 Ph P1 H C8H17 H H H CF3 H	Ī	224	lr	3	0	Ph	P1	Н	Н	Н	C8H17	CF3	Н	Н	Н
	50	225	lr	3	0	Ph	P1	C8H17	н	Н	Н	Н	CF3	Н	Н
227 Ir 3 0 Ph P1 H H C8H17 H H CF3 H H	Ì	226	lr	3	0	Ph	P1	Н	CBH17	Н	Н	Н	^e CF3	Н	Н
		227	lr	3	0	Ph	P1	Н	Н	C8H17	Н	Н	-CF3	Н	Н
55 228 Ir 3 0 Ph P1 H H H C8H17 H CF3 H H		228	Ir	3	0	Ph	P1	Н	н	Н	C8H17	н	+CF3	H	Н
	"	229	lr	3	0	Ph	P1	C8H17	Н	Н	Н	Н	н	CF3	Н
230 Ir 3 0 Ph P1 H C8H17 H H H CF3 H		230	ir	3	0	Ph	P1	Н	C8H17	Н	Н	Н	Н	-CF3	Н

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Table 1-4 (continued)

	No	М	m	n	Α	В	A-R1	A-R2	A-R3	A-R4	B-R5	B-R6	B-R7	B-R8
	231	lr	3	0	Ph	P1	Н	Н	C8H17	Н	Н	Н	CF3	Н
5	₂ 32	lr	3	0	Ph	P1	Н	Н	н	C8H17	Н	Н	°CF3	Н
	233	lr	3	0	Ph	P1	C8H17	. Н	Н	Н	Н	Н	Н	CF3
	234	lr	3	0	Ph	P1	Н	C8H17	н	Н	Н	н	н	CF3
10	235	lr	3	0	Ph	P1	Н	Н	C8H17	н	Н	Н	Н	CF3
	236	ir	3	0	Ph	P1	Н	Н	Н	-C8H17	Н	Н	Н	·CF3
	237	lr	3	0	Ph	P1	F	Н	Н	Н	Н	Н	Н	Н
_	238	lr	3	0	Ph	P1	Н	F	н	Н	Н	Н	Н	Н
15	239	lr	3	0	Ph	P1	Н	Н	F	Н	Н	Н	Н	Н
	240	lr	3	0	Ph	P1	H	Н	Н	F	Н	Н	Н	Н

Table 1-5

							Table 1-	<u>, </u>					
No	М	m	n	Α	В	A-R1	A-R2	A-R3	A-R4	B-R5	B-R8	B-R7	B-R8
241	lr	3	0	Ph	P1	F	F	Н	H	Н	Н	Н	Н
242	lr	3	0	Ph	P1	F	Н	F	Н	Н	Н	Н	H
243	lr	3	0	Ph	P1	Н	F	Н	F	Н	н	Н	Н
244	ir	3	0	Ph	P1	Н	F	F	Н	Н	Н	Н	Н
245	lr	3	0	Ph	P1	Н	F	Н	Н	Н	Н	н	Н
246	lr	3	0	Ph	P1	Н	н	H	F	Н	Н	Н	H
247	lr	3	0	Ph	P1	Н	Н	F	F	н	Н	Н	H.
248	ir	3	0	Ph	P1	F	Н	F	F	Н	Н	Н	Н
249	lr [']	3	0	Ph	P1	F	F	F	Н	н	Н	Н	Н
250	lr	3	0	Ph	P1	Н	F	F	F	Н	Н	Н	Н
251	lr	3	0	Ph	P1	F	F	F	Н	Н	Н	Н	н
252	lr	3	0	Ph	P1	F	F	F	F	Н	н	н	н
253	lr	3	0	Ph	P1	F	Н	Н	Н	СНЗ	Н	Н	Н
254	lr	3	0	Ph	P1	F	Н	н	н	Н	•СН3	Н	Н
255	lr	3	0	Ph	P1	F	Н	Н	Н	Н	Н	СНЗ	Н
256	lr	3	0	Ph	P1	F	н	Н	Н	н	Н	Н	СНЗ
257	lr	3	0	Ph	P1	Н	F	н	Н	СНЗ	Н	Н	Н
258	lr	3	0	Ph	P1	Н	F	Н	Н	Н	СНЗ	Н	Н
259	lr	3	0	Ph	P1	Н	F	Н	Н	Н	Н	СНЗ	Н
260	lr	3	0	Ph	P1	Н	F	Н	Н	Н	Н	Н	•СН3
261	lr	3	0	Ph	P1	Н	Н	F	Н	СНЗ	Н	н	Н
262	lr .	3	0	Ph	P1	Н	Н	F	Н	Н	СНЗ	н	Н
263	ir	3	0	Ph	P1	Н	н	F	н	Н	Н	СНЗ	Н
264	ir	3	0	Ph	P1	н	н	F	Н	н	Н	Н	<i>-</i> СН3
265	lr	3	0	Ph	P1	н	н	Н	F	СНЗ	Н	н	Н
		لــــــا		نـــــا			<u> </u>	.	1		1	1	1

Table 1-5 (continued)

5								TUDIC		milacaj					
267		No	М	m	n	Α	В	A-R1	A-R2	A-R3	A-R4	B-R5	B-R8	B-R7	B-R8
268		266	lr	3	0	Ph	P1	Н	Н	Ξ	F	Н	∙СН3	н	Н:
269 Ir 3 0 Ph P1 F F H H CH3 H H CH3 IF T1 IF T2 IF T3 0 Ph P1 F F H H H CH3 H IF T2 IF T3 0 Ph P1 F F H H H H H CH3 IF T2 IF T3 0 Ph P1 F F H H H H H H CH3 IF T2 IF T3 0 Ph P1 F F H H H H H H CH3 IF T2 IF T3 0 Ph P1 F H F H H H H H CH3 IF T2 IF T3 0 Ph P1 F H F H H H H CH3 IF T2 IF T3 0 Ph P1 F H F H H H H CH3 IF T3 0 Ph P1 F H H F H H H H CH3 IF T3 IF T3 0 Ph P1 F H H F H H H CH3 IF T3 IF T3 0 Ph P1 F H H F H H H CH3 IF T4 T5 T5 T5 T5 T5 T5 T5	<i>5</i>	267	ir	3	0	Ph	P1	Н	н	н	F	Н	н	СНЗ	H
10 270 ir 3 0 Ph P1 F F H H H CH3 H CH3 IF		268	lr	3	0	Ph	P1	Н	Н	Н	F	Н	н	Н	СНЗ
271		269	lr	3	0	Ph	P1	F	F	Н	Н	СНЗ	Н	Н	Н
272	10	270	lr	3	0	Ph	P1	F	F	н	Н	Н	⁻ СН3	Н	Н
273 Ir 3 0 Ph P1 F H F H CH3 H H I I I I I I I I		271	lr	3	0	Ph	P1	F	F	Н	Н	Н	н	СНЗ	Н
274		272	lr	3	0	Ph	P1	F	F	Н	Н	Н	Н	Н	СНЗ
274 Ir 3 0 Ph P1 F H F H H CH3 H 1 275 Ir 3 0 Ph P1 F H F H H H H CH3 I 276 Ir 3 0 Ph P1 F H F H H H H H C 277 Ir 3 0 Ph P1 F H H F H CH3 H H 1 278 Ir 3 0 Ph P1 F H H F H CH3 H H 279 Ir 3 0 Ph P1 F H H F H H CH3 H 1 280 Ir 3 0 Ph P1 F H H F H H CH3 H 1 282 Ir 3 0 Ph P1 H F F H H CH3 H H 282 Ir 3 0 Ph P1 H F F H H CH3 H H 283 Ir 3 0 Ph P1 H F F H H H CH3 H 1 285 Ir 3 0 Ph P1 H F F H H H H CH3 H 286 Ir 3 0 Ph P1 H F F H H CH3 H 1 286 Ir 3 0 Ph P1 H F H F H H CH3 H 290 Ir 3 0 Ph P1 H F H F H H CH3 H 290 Ir 3 0 Ph P1 H H F F H H CH3 H 292 Ir 3 0 Ph P1 H H F F H CH3 H H CH3 40	273	lr	3	0	Ph	P1	F	Н	F	Н	СНЗ	н '	н	Н	
276	15	274	lr	3	0	Ph	P1	F	H	F	Н	Н	-СH3	Н	Н
277		275	lr	3	0	Ph	P1	F	Н	F	н	Н	н	СНЗ	Н
278		276	lr	3	0	Ph	P1	F	Н	F	Н	н "	Н	Н	СНЗ
279 Ir 3 0 Ph P1 F H H F H H CH3 I 280 Ir 3 0 Ph P1 F H H F H H H CH3 I 281 Ir 3 0 Ph P1 H F F H CH3 H H CH3 I 282 Ir 3 0 Ph P1 H F F H H CH3 H I 283 Ir 3 0 Ph P1 H F F H H H CH3 I 285 Ir 3 0 Ph P1 H F F H H H H CH3 I 286 Ir 3 0 Ph P1 H F H F H CH3 H I 286 Ir 3 0 Ph P1 H F H F H CH3 H I 286 Ir 3 0 Ph P1 H F H F H H H CH3 I 288 Ir 3 0 Ph P1 H F H F H H H CH3 I 289 Ir 3 0 Ph P1 H H F F CH3 H H CH3 I 290 Ir 3 0 Ph P1 H H F F H CH3 H CH3 I 292 Ir 3 0 Ph P1 H H F F H H H CH3 I 293 Ir 3 0 Ph P1 H H F F H H H CH3 20	277	lr	3	0	Ph	P1	F	H	Н	F	СНЗ	Н	Н	Н	
280		278	lr.	3	0	Ph	P1	F	Н	Н	F	Н	СНЗ	Н	Н
281 Ir 3 0 Ph P1 H F F H CH3 H H Ir 282 Ir 3 0 Ph P1 H F F H H CH3 H Ir 283 Ir 3 0 Ph P1 H F F H H H CH3 Ir 284 Ir 3 0 Ph P1 H F F H H H H CH3 Ir 285 Ir 3 0 Ph P1 H F F H H H H H CH3 Ir 286 Ir 3 0 Ph P1 H F H F H CH3 H Ir 287 Ir 3 0 Ph P1 H F H F H CH3 H Ir 288 Ir 3 0 Ph P1 H F H F H H H CH3 Ir 289 Ir 3 0 Ph P1 H H F F CH3 H H CH3 Ir 290 Ir 3 0 Ph P1 H H F F H CH3 H Ir 291 Ir 3 0 Ph P1 H H F F H H H CH3 Ir 292 Ir 3 0 Ph P1 F F F H H H CH3 H 293 Ir 3 0 Ph P1 F F F H H H CH3 H 294 Ir 3 0 Ph P1 F F F H H H CH3 H 295 Ir 3 0 Ph P1 F F F H H H H CH3 296 Ir 3 0 Ph P1 F F F H H H H H CH3 297 Ir 3 0 Ph P1 F F F H H H H H CH3 298 Ir 3 0 Ph P1 F F F H H H H H CH3 299 Ir 3 0 Ph P1 F F F H H H H H CH3 290 Ir 3 0 Ph P1 F F F H H H H H CH3 291 Ir 3 0 Ph P1 F F F H H H H H CH3 291 Ir 3 0 Ph P1 F F F H F H CH3 H 291 Ir 3 0 Ph P1 F F F H F H CH3 H 291 Ir 3 0 Ph P1 F F F H F H CH3 H 291 Ir 3 0 Ph P1 F F F H F H CH3 H 291 Ir 3 0 Ph P1 F F F H F H CH3 H 291 Ir 3 0 Ph P1 F F F H F H CH3 H 291 Ir 3 0 Ph P1 F F F H F H CH3 H 291 Ir 3 0 Ph P1 F F F H F H CH3 H 291 Ir 3 0 Ph P1 F F F H F H CH3 H 291 Ir 3 0 Ph P1 F F F H F H CH3 H 291 Ir 3 0 Ph P1 F		279	lr	3	0	Ph	P1	F	Н	Н	F	Н	н	СНЗ	Н
281	25	280	lr	3	0	Ph	P1	F	Η	Ξ	F	н	Η	Н	•СН3
283 Ir 3 0 Ph P1 H F F H H H CH3 I 284 Ir 3 0 Ph P1 H F F H H H H CH3 I 285 Ir 3 0 Ph P1 H F H F CH3 H H I 286 Ir 3 0 Ph P1 H F H F H CH3 H I 287 Ir 3 0 Ph P1 H F H F H H CH3 I 288 Ir 3 0 Ph P1 H F H F H H H CH3 I 289 Ir 3 0 Ph P1 H H F F CH3 H H I 290 Ir 3 0 Ph P1 H H F F H CH3 H I 291 Ir 3 0 Ph P1 H H F F H H CH3 H 292 Ir 3 0 Ph P1 F F F H H H CH3 H 293 Ir 3 0 Ph P1 F F F H H H CH3 H 294 Ir 3 0 Ph P1 F F F H H H CH3 H 296 Ir 3 0 Ph P1 F F F H H H H CH3 298 Ir 3 0 Ph P1 F F F H H H H CH3 299 Ir 3 0 Ph P1 F F F H H H H CH3 290 Ir 3 0 Ph P1 F F F H H H H CH3 291 Ir 3 0 Ph P1 F F F H H H H CH3 291 Ir 3 0 Ph P1 F F F H F H CH3 H 300 P1 P1 F F F F H F H CH3 H 300 P1 P1 F F F F H F H CH3 H 300 P1 P1 F F F F H F H CH3 H 300 P1 P1 F F F F H F H CH3 H 300 P1 P1 F F F F H F H CH3 H 300 P1 P1 F F F F H F H CH3 H 300 P1 P1 F F F F H F H CH3 H 300 P1 P1 F F F F H F H CH3 H 300 P1 P1 F F F F H F H CH3 H 300 P1 P1 F F F F H F H CH3 H 300 P1 P1 F F F F H F H CH3 H 300 P1 P1 F F F F H F H CH3 H 300 P1 P1 F F F F H F H CH3 H 300 P1 P1 F F F F H F H CH3 H 300 P1 P1 F F F F H F H CH3 H 300 P1 P1 F F F F H F H CH3 H 300 P1 P1 F F F F H F	25	281	Ir	3	0	Ph	P1	Н	F	F	Н	СНЗ	Н	Н	н
284 Ir 3 0 Ph P1 H F F H H H H . C 285 Ir 3 0 Ph P1 H F H F CH3 H H I 286 Ir 3 0 Ph P1 H F H F H CH3 H I 287 Ir 3 0 Ph P1 H F H F H H H CH3 I 288 Ir 3 0 Ph P1 H F H F H H H C 289 Ir 3 0 Ph P1 H H F F CH3 H H I 290 Ir 3 0 Ph P1 H H F F H CH3 H I 291 Ir 3 0 Ph P1 H H F F H H H C 292 Ir 3 0 Ph P1 H H F F H H H C 293 Ir 3 0 Ph P1 F F F H H C 45 294 Ir 3 0 Ph P1 F F F H H H C 296 Ir 3 0 Ph P1 F F F H H H H C 297 Ir 3 0 Ph P1 F F F H H H H C 298 Ir 3 0 Ph P1 F F F H H H H H C 298 Ir 3 0 Ph P1 F F F H F H C 30 C C C C C C C 30 C C C C C C 30 C C C C C C 40 C C C C C 41 C C C C C 42 C C C C C 43 C C C C C 44 C C C C 45 C C C C 46 C C C C 47 C C C C 48 C C C C 49 C C C C 40 C C C C 40 C C C C 40 C C C C 41 C C C C 42 C C C C 43 C C C C 44 C C C C 45 C C C C 46 C C C C 47 C C C C 48 C C C C 49 C C C C 40 C C C C 41 C C C C 42 C C C C 43 C C C C 44 C C C C 45 C C C 46 C C C C 47 C C C C 48 C C C C 49 C C C C 40 C C C C 40 C C C 40 C C C C 40 C C C C 41 C C C 42 C C C C 43 C C C C 44 C C C C 45 C C C C 46 C C C C 47 C C C C		282	lr	3	0	Ph	P1	H,	F	F	Н	Н	∘СНЗ	Н	Н
285 Ir 3 0 Ph P1 H F H F CH3 H H CH3 C		283	ir	3	0	Ph	P1	H	F	F	Н	Н	Н	СНЗ	. н
286 3 0 Ph P1 H F H F H	30	284	ir	3	0	Ph	P1	H	F	F	Н	Н	Н	н.	-СН3
287		285	lr	3	0	Ph	P1	Н	F	Н	F	СНЗ	Н	Н	Н
288 3 0 Ph P1 H F H F H H C C C C C C		286	lr	3	0	Ph	P1	H	F	Н	F	Н	СНЗ	Н	Н
288 3 0	35	287	lr	3	0	Ph	P1	Н	F	Н	F	Н	Н	СНЗ	Н
290		288	Ir	3	0	Ph	P1	Н	F	Н	F	Н	Н	н	СНЗ
40		289	lr	3	0	Ph	P1	H	H	F	F	СНЗ	Н	Н	Н
292 ir 3 0 Ph P1 H H F F H H H C 293 ir 3 0 Ph P1 F F F H CH3 H H 294 ir 3 0 Ph P1 F F F H H CH3 H 295 ir 3 0 Ph P1 F F F H H H CH3 296 ir 3 0 Ph P1 F F F H H H CH3 297 ir 3 0 Ph P1 F F F H H H H CH3 298 ir 3 0 Ph P1 F F F H H CH3 H H 299 ir 3 0 Ph P1 F F F H F CH3 H H		290	lr	3	0	Ph	P1	Н	I	F	F	Н	СНЗ	Н	н
293 Ir 3 0 Ph P1 F F F H CH3 H CH3 H CH3 H CH3	40	291	ir	3	0	Ph	P1	Н	Н	F	F	Н	Н	СНЗ	Н
294 Ir 3 0 Ph P1 F F F H H CH3 H 295 Ir 3 0 Ph P1 F F F H H H CH3 296 Ir 3 0 Ph P1 F F F H H H H C 291 Ir 3 0 Ph 5P1 F F H F CH3 H H 50 298 Ir 3 0 Ph P1 F F H F H CH3 H		292	lr	3	0	Ph	P1	н	Н	F	F	н	H	Ξ	СНЗ
295 Ir 3 0 Ph P1 F F F H H H CH3 C		293	lr	3	0	Ph	P1	F	F	F	H	СНЗ	н	Н	н -
295 Ir 3 0 Ph P1 F F F H H H CH3 C	45	294	lr	3	0	Ph	P1	F	F	F	H	Н	-CH3 [®]	Н	Н
291 Ir 3 0 Ph 5P1 F F H F CH3 H H F CH3 H H S CH3 H CH	1	295	lr	3	0	Ph	P1	F	F	F	Η	Н "	н	СНЗ	H
50 298 Ir 3 0 Ph P1 F F H F H CH3 H		296	lr	3	0	Ph	P1	F	F	F	Н	Н	H	Н	СНЗ
	:	291	lr	3	0	Ph	5P1	F	F	Н	F	СНЗ	Н	Н	Н
200 k 3 0 Pb Pt E E H E H CIO	50	298	lr	3	0	Ph	P1	F	F	Н	F	Н	снз	Н	Н
299 3 0 FII FI F F H F H CH3		299	ir	3	0	Ph	P1	F	F	Н	F	н	н	СНЗ	Н
300 Ir 3 0 Ph P1 F F H F H H H +C		300	Ir	3	0	Ph	P1	F	F	н	F	н	н	н	-СН3

Table 1-6

								Table					- 4	
	No	М	m	n	Α	В	A-R1	A-R2	A-R3	A-R4	B-R5	B-R6	B-R7	B-R8
5	301	lr	3	0	Ph	P1	F	н	F	F	СНЗ	Н	H	Н
	302	lr	3	0	Ph	P1	F	Н	F	F	Н	СНЗ	Н	Н
	303	Ir	3	0	Ph	P1	F	Н	F	F	н	Н	CH3 [^]	Н
	304	lr	3	0	Ph	P1 .	F	Н	F	F	Н	Н	Н	СНЗ
10	305	lr	3	0	Ph	P1	F	F	F	F	СНЗ	Н	Н	Н
	306	lr	3	0	Ph	P1	F	F	F	F	Н	СНЗ	Н	Н
	307	lr	3	.0	Ph	P1	F	F	F	F	Н	Н	-СН3	Н
15	308	lr	3	0	Ph	P1	F	F	F	F	Н	Н	Н	СНЗ
	309	lr	3	0_	Ph	P1	CF3	Н	Н	Н	Н	Н	н *	Н
	310	lr	3	0	Ph	Pt	Н	CF3	Н	н	Н	Н	Η	н
	311	lr	3	0	Ph	P1	ı	Н	CF3	Н	Н	н	Ή	Н
20	312	lr	3	0	Ph	P1	ı	CF3	H	CF3	н	Н	Н	Н
	313	lr	3	0	Ph	P1	CF3	CF3	Н	Н	Н	н	н	Н
	314	lr	3	0	Ph	P1	CF3	H	CF3	Ή	Η	Ι	Ŧ	Н
25	315	lr	3	0	Ph	P1	CF3	H	H	CF3	Н	Н	Н	Н
	316	lr	3	0	Ph	P1	н	CF3	CF3	Н	н	Н	Н	Н
	317	lr	3	0	Ph	P1	H	Ξ	C3F7C2H4	Η	H	Η	Н	н.
20	318	lr	3	0	Ph	P1	I	Ξ	C7F15	Н	Н	Н	Н	Н
30	316	lr	3	0	Ph	P1	H	Η	CF3	CF3	Ι	Н	Н	Н
	320	lr	3	0	Ph	P1	CF3	Ξ	CF3	CF3	Н	K	Н	Н
	321	lr	3	0	Ph	P1	CF3	CF3	CF3	Н	Н	Н	М	Н
35	322	lr	3	0	Ph	P1	Н	CF3	CF3	·CF3	Н	Н	Н	Н
	323	lr	3	0	Ph	P1	CF3	CF3	CF3	Н	H	Ŧ	Н	Н
	324	lr	3	0	Ph	P1	CF3	CF3	CF3	CF3	Н	Н	н	Н
40	325	lr	3	0	Ph	P1	CF3	Н	Н	Н	СНЗ	Н	Н	Н
40	326	lr	3	0	Ph	P1	CF3	Н	Н	Н	Н	СНЗ	Н	Η ,
	327	lr	3	0	Ph	P1	CF3	Ŧ	H	Н	Н	Н	СНЗ	Н
:	328	lr	3	0	Ph	P1	CF3	Ħ	H	Н	Н	Н	Н	СНЗ
45	329	Ir	3	0	Ph	P1	I	CF3	Η	Н	СНЗ	Н	Н	н
	330	lr	3	0	Ph	P1	Τ	CF3	Н	Н	Н	СНЗ	Н	Н
	331	lr	3	0	Ph	P1	H	CF3	н	Н	Н	Н	СНЗ	Н
50	332	ir	3	0	Ph	P1	Н	Н	Н	Н	Н	Н	Н	-СН3
30	333	lr	3	0	Ph	P1	Н	н	CF3	Н.	СНЗ	Н	Н	Н
	334	lr	3	0	Ph	P1	Н	Н	CF3	Н	н	€Н3	н	Н
	335	lr	3	0	Ph	P1	Н	н	CF3	Н	Н	Н	СНЗ	Н
55	336	lr	3	0	Ph	P1	H	Н	CF3	Н	н	Н	Н	СНЗ
	337	lr	3	0	Ph	P1	Н	Н	н	-CF3	² СН3	Н	Н	Н
											*			

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Table 1-6 (continued)

	No	М	т	n	Α	В	A-R1	A-R2	A-R3	A-R4	B-R5	B-R6	B-R7	B-R8
	338	lr	3	0	Ph	P1	Н	н	Н	CF3	Η	СНЗ	Н	Н
5	339	lr .	3	0	Ph	P1	Н	н	Н	CF3	н	н	СНЗ	Н
	340	lr	3	0	Ph	P1	Н	Н	Н	CF3	Н	Н	н	СН3
	341	lr	3	0	Ph	P1	CF3	CF3	Н	Н	СНЗ	Н	Н	Н
10	342	lr	3	0	Ph	P1	CF3	CF3	Н	Н	н	СНЗ	н	Н
	343	lr	3	0	Ph	P1	CF3	CF3	Н	Н	Н	Н	СНЗ	Н
	344	lr	3	0	Ph	P1	CF3	CF3	Н	Н	Н	Н	Н	СН3
	345	lr	3	0	Ph	P1	CF3	Н	CF3	Н	СНЗ	Н	Н	Н
15	346	lr	3	0	Ph	P1	CF3	·H	CF3	Н	н	СНЗ	H	Н
	347	lr	3	0	Ph	P1	CF3	H	CF3	Н	н	Н	СНЗ	Н
	348	lr	3	0	Ph	P1	CF3	Н	CF3	Н	Н	Н	Ħ	СНЗ
20	349	lr	3	0	Ph	P1	CF3	H	Н	CF3	СНЗ	Н	Н	н
	350	lr	3	0	Ph	P1	CF3	Η	Н	CF3	Н	СНЗ	Н	Н
	351	lr	3	0	Ph	P1	CF3	Ι	Н	CF3	Н	н	- СН3	Н
25	352	lr	3	0	Ph	P1	CF3	Ξ	Н	CF3	Н	Н	Н.	СНЗ
25	353	lr	3	0	Ph	P1	Н	CF3	CF3	Н	-CH3	н	Н	Н
	354	lr	3	0	Ph	P1	Н	CF3	CF3	Н	Н	СНЗ	Н	Н
	355	lr	3	0	Ph	P1	Н	CF3	CF3	Н	Н	Н	СНЗ	Н
30	356	lr	3	0	Ph	P1	H	CF3	CF3	Н	Н	Н	Н	СНЗ
	357	lr	3	0	Ph	P1	H	ÇF3	Н	CF3	СНЗ	Н	Н	Н
	358	lr	3	0	Ph	P1	Н	CF3	Н	CF3	Н	СНЗ	Н	Н
35	359	lr	-3	0	Ph	P1	Н	CF3	Н	CF3	Н	Н	СНЗ	Н
33	360	lr	3	0	Ph	P1	Н	CF3	Н	CF3	Н	Н	Н	СНЗ

Table 1-7

40	No	М	m	n	Α	В	A-R1	A-R2	A-R3	A-R4	B-R5	B-R6	B-R7	B-R8
	361	lr	3	0	Ph	P1	н	Н	CF3	CF3	СНЗ	H	Н	Н
	392	lr	3	0	Ph	P1	Н	Н	CF3	CF3	Н	СНЗ	Н	Н
45	383	lr	3	0	Ph	P1	Н	Н	CF3	CF3	Н	Н	СНЗ	Н
	364	lr	3	0	Ph	P1	Н	Н	CF3	CF3	Н	Н	Н	•СН3
	365	lr	3	0	Ph	P1	CF3	CF3	CF3	Н	СНЗ	Н	н	Н
	368	lr	3	0	Ph	P1	CF3	CF3	CF3	Н	н	СНЗ	Н	Н
50	367	Ir	3	0	Ph	P1	CF3	CF3	CF3	Н	Н	Н	СНЗ	Н
	368	lr	3	0	Ph	P1	CF3	CF3	CF3	Н	Н	Н	Н	СНЗ
	369	Ir	3	0	Ph	P1	CF3	CF3	Н	CF3	СНЗ	Н	н	Н
55	370	lr	3	0	Ph	P1	CF3	CF3	Н	CF3	Н "	СНЗ	Н	Н
	371	lr	3	0	Ph	P1	CF3	CF3	Н	CF3	Н	Н,	СНЗ	Н
	372	lr	3	0	Ph	P1	CF3	CF3	Н	CF3	Н	Н	Н	_г СН3

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Table 1-7 (continued)

	No	М	m	n	Α	В	A-R1	A-R2	A-R3	A-R4	B-R5 ¹	B-R6	B-R7	B-R8
	373	ir	3	0	Ph	P1	CF3	Н	CF3	CF3	СНЗ	Н	Н	Н
5	374	lr	3	0	Ph	P1	CF3	Н	CF3	CF3	Н	СНЗ	H	Н
	375	lr	3	0	Ph	P1	CF3	Н	CF3	CF3	Н	Н	СНЗ	Н
	376	ir	3	0	Ph	P1	CF3	н	CF3	CF3	Н	Н	Η	СН3
10	377	ir	3	0	Ph	P1	·CF3	CF3	CF3	CF3	СНЗ	Н	Н	H
	37B	lr	3	0	Ph	P1	CF3	CF3	CF3	-CF3	Н	CH3	Н	Н
	379	lr	3	0	Ph	P1	CF3	CF3	CF3	CF3	Н	Н	СНЗ	Н
	380	lr	3	0	Ph	P1	CF3	CF3	CF1	CF3	Ξ	Н	Н	°СН3
15	381	lr	3	0	Ph	P1	F	CF3	H	Ħ	Ξ	Н	Н	Н
	3BZ	lr	3	0	Ph	P1	F	CF3	I	CF3	I	Ξ	H,	Н
	383	Ir	3	0	Ph	P1	F	Ι	I	CF3	I	Ξ	H	Н
20	384	lr	3	0	Ph	P1	Ξ	CF3	F	Η	I	Η	Η	Н
	385	lr	3	0	Ph	P1	Н	CF3	F	CF3	Н	Н	Н	Н
	386	lr	3	0	Ph	P1	Н	H ,	F	CF3	Н	H	H	Н
25	387	lr	3	0	Ph	P1	F	CF3	F	Н	Н	Н	H	Н
25	388	Ir	3	0	Ph	P1	F	Н	F	CF3	Ŧ	Н	Н	Н
	389	lr	3	0	Ph	P1	Н	СНЗ	F	Н	Н	Н	Н	Н
	390	lr	3	0	Ph	P1	Н	СНЗ	CF3	Н	Η "	Н	Н	Н
30	39f	lr	3	0	Ph	P1	F	CF3	Н	CF3	Н	Н	Н	Н
	392	lr	3	0	Ph	P1	CF3	Н	F	Н	Н	Н	Н	Н
	393	ir	3	0	Ph	P1	Н	CF3	F	Н	Н	СНЗ	Н	Н
35	394	lr	3	0	Ph	P1	Н	CF3	F	CF3	н	СНЗ	Н	Н
55	395	lr	3	0	Ph	P1	Н	Н	F	CF3	Н	СНЗ	Н	Н
	396	lr	3	0	Ph	P1	F	CF3	F	Н	Н	СНЗ	Н	Н :
	397	lr	3	0	Ph	P1	F	Н	F	CF3	Н	СНЗ	н	Н
40	398	lr _	3	0	Ph	P1	F	CF3	F	CF3	Н	СНЗ	Н	Н
	399	lr	3	0	Ph	P1	F	CF3	Н	Н	Н	Н	-CH3	Н
	400	Ir	3	0	Ph	P1	F	CF3	Н	CF3	Н	Н	СНЗ	Н
45	401	Ir	3	0	Ph	P1	F	Н	Н	CF3	Н	Н	СНЗ	Н
	402	Ir	3	0	Ph	P1	Н	CF3	F	+	Н	Н	СНЗ	Н
	403	lr	3	0	Ph	P1	Н	CF3	F	CF3	Н	Н	СНЗ	Н
	404	lr	3	0	Ph	P1	Н	Н	F	CF3	Н	Н	CH3	Н
50	405	lr	3	0	Ph	P1	F	°CF3	F	Н	Н	Н	СНЗ	Н
	406	lr	3	0	Ph	P1	F	Н	F	CF3	Н	Н	СНЗ	Н
	407	lr	3	0	Ph	P1	F	CF3	F	∻CF3	н	Н	[∞] CH3	Н
55	406	lr_	3	0	Ph	P1	F	I	Ι	Н	Н	CF3	н	Н ,
	409	lr	3	0	Ph	P1	Н	F	I	H	Н	CF3	н	Н
	410	lr	3	0	Ph	P1	н	Н	F	Н	Н	CF3	н	Н

Table 1-7 .(continued)

No	М	m	n	Α	В	A-R1	A-R2	A-R3	A-R4	B-R5	B-R6	8-R7	B-R8
411	lr	3	0	Ph	P1	Н	Н	Н	F	Н	CF3	Н	Н
412	lr	3	0	Ph	P1	F	Н	Н	Н	н	Н	CF3	Н
413	lr	3	0	Ph	P1	Н	F	Н	Н	Н	Н	·CF3	Н
414	lr	3	0	Ph	P1	Н	H	F	Н	н	Н	CF3	Н
415	lr	3	0	Ph	P1	Н	Н	Н	F	Н	Н	CF3	Н
418	ir	3	0	Ph	P1	Н	F	Н	F	Н	CF3	Н	Н
417	lr	3	0	Ph	P1	Н	F	Н	F	Н	CF3	Н	Н
418	lr	3	0	Ph	P1	Н	F	Н	F	Н	CF3	Н	Н
419	Ir	3	0	Ph	P1	Н	F	Н	F	Н	°CF3	Н	Н
420	lr	3	0	Ph	P1	Н	F	Н	F	Н	Н	·CF3	Н

5		B-R8	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	Ξ	I	I	Σ	I	Τ	I
		B-R7	CF3	CF3	CF3	I	I	I	I	CF3	CF3	CF3	CF3	I	I	I	I	CF3	CF3	CF3	CF3	I	I	I	I	I	I
10		B-R6	I	I	I	CF3	CF3	CF3	CF3	I	I	I	ı.	CF3	CF3	CF3	CF3	I	I	I	I	I	I	I	Η.	I	I
15		B-R5	Ξ	I	Ξ	Ξ	Ξ	I	I	I	I	I	I	Ι.	I	H	I	I	I	I	π	Ξ	I	I	I	I	I
		A-R4	F	ъ	ш	I	I	I	CF3	I	I	I	CF3	I	I	н	CF3	н	Ŧ	I	CF3	I,	I	H	СНЗ	I,	I
20		A-R3	I	H	I	I	I	CF3	I	ı	I	CF3	Ι.	CF3	CF3	CF3	I	CF3	CF3	CF3	I	I	Œ.	СНЗ	T	н	I
25		A-R2	F	F	F	I	CF3	н	I	I	CF3	н	I	н	ш	I	I	H	F	I	H	I.	윉	I	I	I	C2H5
30	Table 1-8	A-R1	н	H	I	CF3	н	н	н	CF3	H	н	Ι	CF3	I	CF3	I	CF3	I	CF3	Ι	양	I	I	I	C2H5	I
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35		ח		-	•			•	•	•	•	•	•	•								I	I	I	I	Œ	I
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55		ž	421	422	423	424	425	426	427	428	429	430	431	432	433	434	435	436	437	438	439	440	144	445	443	4	445

5		B-R8	I	I	I	I	I	I	I	I	I	I	Ι	Ι	I	I	I	I	I	I	I	I	r	I	I	I	I
		B-R7	I	I	I	Ξ	Ι.	ı	I	I	I	I	I.	r	I.	Τ	Ξ	I	Ŧ	I	Ξ	I	I	I	I	I	I
10	•	B-R6	I	I	Ŧ	I	Ŧ	ェ	Ŧ	Ŧ	I	I.	Ŧ	I	H	H	Ŧ	н	Ŧ	I	I	I	I	Ι	Τ	I	I
15		B-R5	Ξ	н	I	I	I	I	I	I	I	I	I	I	Ι	Ι	I	I	Τ	I.	Τ	н	ΞE	I	I	Œ	I
		A-R4	I	C2H5	I	I	I,	C3H7	I	I	I	C4H9	н	H	Н	С6Н13	н	I	H	C8H17	I	I	I	C12H25	H	I	I
20		A-R3	C2H5	I	I	I	СЗН7	Ŧ	I	I	C4H9	I	I	н	С6Н13	I	I	I	C8H17	H	I	I	C12H25	I	I	I	C15H31
25	(pər	A-R2	н	I	Ι	C3H7	I	н	н	C4H9	I	H	I	C6H13	н	I.	I	C8H17	I	Ι.	н	C12H25	I	I	I	C15H31	I
30	-8 (continued)	A-R1	н	I	C3H7	I	I	H	С4Н9	I	I	I	С6Н13	-	I	Ξ,	CBH17	I	I	н	C12H25	I	I	I	C15H31	I	I
	Table 1-8	5	СНЗ	왕	CH3	윉	CH3	CH3	СНЗ	당	CH3	당	몽	CH3	CH3	당	윉	CH3	CH3	CH3	CH3	CH3	CH3	器	윉	CH3	양
35		ſ	н	I	I	I	I	I.	I,	I	I,	I,	I	I	I	I	I	I	I	I	I	I	I	Œ	I	I	I
40		Е	СНЗ	СНЗ	СНЗ	СНЗ	CH3	СНЗ	CH3	CH3	문 문		뜐	왕	CH3	똜	托	왕	뚱	쭝	뚱	왕	뚱	뚱	뚱	뚱	뚱
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5	B-R8	I	I	I	I	I	τ	I	I	I	I
	B-R7	н	r	I	I	I	I	I	I	I	I
10	B-R6	I	I	I	I	I	Ŧ	Ŧ	I	ı	I
15	B-R5	I	I	I	I	I	I	н	I	H	ı
	A-R4	C15H31	I	I	I	I	Ι	I	I	I	н
20	A-R3	Ξ	I	I	I	I	I	н	I	н	H
25 5	A-R2	I	I	I	I	I	I	I	I	I	СНЗ
30	A-R1	I	I	I	I	I	I	I	I	I	I
F	G G	왕	CF3	CH3	CF3	CF3	윉	C4H9	CH3	СНЗ	СНЗ
35	٦	I	윉	ц	СНЗ	ш	CF3	ц	C2HS	C4H9	СНЗ
40	Е	윉	당	CH3	CF3	CF3	땑	C4H9	СНЗ	СНЗ	СНЗ
	В	P1	Pŧ	P1	7	P1	7	Ы	P	P1	P1
45	٨	Ъ	μ	чЬ	Ph	Ph	띰	Ph	F.	Ph	Ph
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55	ş	471	472	473	474	475	476	477	478	479	480

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		B-R8	I	I	Ξ	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	Ι	Ι	Ι	I
10		8-R7	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	Ι	I	I
		B-R6	Ξ	I	I	I	I	I	I	I	I	I	I	Ξ	I	I	I	I	ı,	I	H	I	I	Ξ	СНЗ	СНЗ	СНЗ
15		B-R5	I	I	I	I	Ξ	I	I	I	I	Ξ	Ξ	I	I	I	I.	I	н	I	I	I	I	I	I	I	I
20		A-R4	I	I	Ι	Ι	I	Ξ	Ŧ	I	I	Ι	I	Ξ	I	I	I	I	I	I	H	I	I	I	Ħ	I	I
		A-R3	I	ェ	Ξ	I	Ī	Ξ	ェ	I	ш.	I	I	I	I	ı,	I	Ξ	I	I	I	I	I	I	Œ	I	I
25		A-R2	CH3	윉	器	托	몽	윉	ш	u	I	ш	ш	ட	F	LL.	SF3	CF3	CF3	CF3	CF3	CF3	CF3	CF3	I	I	I
	6-	A-A1	I	I	I	I	I	I	I	I	I	I	I	I	Η·	I	I	Ι	I	I	I	r	I	I	I	I	I
30	Table 1-9	_o	СНЗ	CF3	CF3	СНЗ	C4H9	СНЗ	СНЗ	СНЗ	СНЗ	CF3	CF3	CH3	C4H9	СНЗ	SH3	СНЗ	СНЗ	CF3	CF3	뚱	С4Н9	CH3	ĊНЗ	СНЗ	왕
35		٦	T.	СНЗ	F	CF3	L.	C2H5	Ŧ	СНЗ	I	СНЗ	ш	CF3	ш	C2H5	ı	СНЗ	ı.	CH3	F	CF3	u.	C2H5	H	СНЗ	1
		ш	CH3	CF3	CF3	СНЗ	C4H9	СНЗ	托	СНЗ	СНЗ	CF3	CF3	СНЗ	C4H9	СНЗ	CH3	СНЗ	СНЗ	CF3	CF3	뜐	C4H9	CH3	СНЗ	СНЗ	CH3
40		8	P1	P1	Ы	P1	E	E	<u>.</u>	P1	P1	P1	P1	P1	P1	P1	P1	P1 (P1	P	P1	٥	<u>P</u>	<u>P</u>	P1 (P1 (P .
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10		B-R7	I	I	I	I	I	SH3	CH3	CH3	CH3	CH3	윉	SHS	CH3	I	I	I	I	I.	I	I	I	CH3	CH3	
		B-R6	CH3	CH3	몽	몽	몽	I	I	I	I	I	I	I	I	윉	왕	문	CH3	CH3	CH3	CH3	윉	I	I	
15		B-R5	I	I	I	I	I	I	I	I	Ξ	I	I	I	Ι	Ξ	I	I	I	I	I	I	I	I	Ŧ	-
20		A-R4	Ι	Ι	I	Ξ	I	I	Ξ	I	Ι	I	I	I	I	I	Œ	Ξ	I	I	I	I	I	I	Ξ	;
		A-R3	I	I	Ξ	Ξ	I	I	I	I	I	I	Ξ	I	I	ட	ட	ட	ц	u.	ш	ш	ш	u.	ъ	-
25	ਰੇ	A-R2	I	I	I	I	I	Ξ	I	I	I	I	I.	I	I '	CF3	CF3	CF3	CF3	CF3	CF3	CF3	CF3	3	I	[-
	(continued)	A-A1	I	I	I	I	I	Ξ	I	I	I	I	I	I	I	I	I	I	I	I	I	I	I	ш	ш	
30	Table 1-9 (G	CF3	CF3	托	C4H9	윉	CH3	뚱	왕	CF3	CF3	뜐	C4H9	CH3	당	윉	CH3	CF3	CF3	윉	C4H9	CH3	CH3	CH3.	
35	Tat	٦	СНЗ	ш	CF3	ıц	C2H5	I	CH3	IL.	윉	ш	CF3	L.	C2H5	I	CH3	ъ.;	СНЗ	ш.	CF3	u.	C2H5	I	CH3	
	:	Е	CF3	CF3	СНЗ	C4H9	윉	CH3	СНЗ	СНЗ	CF3	CF3	당	C4H9	CH3	СНЗ	CH3	당	CF3	CF3	CH3	C4H9	CH3	CH3	СНЗ	
40		В	P1	P1	P1	P1	ы	7	P1	Ы	P1	Ε	F	ы	۳.	P1	£	£	P1	P	ā.	٦	٦	E	F.	 -
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15	B-R5	I	ı	н	Н	I
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	8-8	Ŧ	I	Ξ	ェ	I	Ξ	Ŧ	I	Ξ	Ξ
	B-R7	CH3	СНЗ	댅	CH3	I	I	I	CH3	CF3	CF3
	B-R6	Ŧ	I	I	I	I	I	I	I	I	I
	B-R5	I	Ξ	I	I	I	I	I	I	I	I
	A-R4	Ξ	Ξ	I	I	ıτ	ட	L	ш	I	I
	A-R3	ъ.	ш	ш	ш	I	I	Ξ	Σ	u.	ш
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Table 1-9 (continued)	5	CF3	당	C4 H9	托	윉	SH3	왕	CF3	CF3	托
Tat	ſ	F	CF3	ш	C2H5	I	윉	ш	윉	ட	CF3
	Ε	CF3	СНЗ	C4H9	윉	윉	당	윉	SE3	CF3	CH3
	В	P1	P1	2	2	P	F	P1	P	14	ᇤ
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	Σ	-	ır	Ir	ır	Ir	II.	11	İr	JI.	<u>-</u>
	Š	531	532	533	S34	535	538	232	538	539	540

5		B-R8	ı	I	I	I	I	I	I	I	I	I	I	I	I	I	н	I	I	I	I	Ŧ	I	I	I	I	I
10		B-R7	CF3	CF3	I	I	I	I	I	I	I	I	I	ı	I	Н	I	I	I	н	CH3	СНЗ	СНЗ	СНЗ	СНЗ	СНЗ	CH3
		B-R8	ェ	I	CF3	CF3	CP3	SF3	SF3	CF3	CF3	CF3	I	I	I	I	Ŧ	I	I	I	I	I	I	Ι	I	I	I
15		B-R5	±	I	Н	I	H	I	I	Ξ	н	н	Н	н	н	H	н	H	н	н	Н	I	I	Η	I	Ŧ	I
		A-R4	I	н	н	I	I	ı	I	I	I	н	ı.	I.	I	I	I	I	I	I	н	H	H,	Ή	I	H	I
20		A-R3	F	F	ıĿ	ட	ш	Ħ	ш	ш	ш	ш	ıL	ц.	F	ıĿ	ъ.	ч	L.	ш,	ч.	ш.	F	L.	F	F	ш
25		A-R2	I	I	I	I	r	I	I	I	I	I (CF3	CF3	CF3	СРЗ	CF3	CF3	CF3	CP3	CF3	CF3	CF3	CF3	CF3	CF3	CF3
20		A-R1	Ξ	I	I	I	I	I	I	I	Ι	Ι	I	I	Ι	I	I	I	I	I	I	I	I	I	I	I	I
30	Table 1-10	ပ	C4H9	СНЗ	СНЗ	CH3	CH3	CF3	CF3	СНЗ	C4H9	CH3	CH3	CH3	CH3	CF3	CF3	СНЗ	C4H9	CH3	당.	CH3	당.	СРЗ	CF3	СНЗ	C4H9
	_	ſ	ш	C2H5	I	СНЗ	ч	СНЗ	F	CF3	ш	C2H5	I	CH3	Н	СНЗ	ı	CF3	u.	C2H5	I	왕	ш	왕	T.	CF3	F
35		ш	C4H9	СНЗ	СНЗ	СНЗ	СНЗ	CF3	CF3	СНЗ	C4H9	СНЗ	양	CH3	СНЗ	CF3	CF3	СНЗ	C4H9	뚬.	왕	당	籽	CF3	CF3	СНЗ	C4H9
40		B'orB				•			•	•	•	•		•	•	•			,		,						
		ω	P1	7	P1	P1	P1	Ρł	P1	Ьł	Ρ1	P1	Ы	Ы	P1	ŀd	l ld	Ы	2	2	<u>a</u>	2	<u>e</u>	4	Ε.	g.	F.
45		٧	Ph		Ph	Ph	Ph	P.	Æ	٩	P.	€	ď	£	£	F.	둡	£	€	A.	Ph						
		c	1	1 Ph	-	1	1	-	-	1	1	1	+	-	-	-	-	-	_	_	_	_	_	_	_	-	-
50		ε	2	2	7	2	2	7	2	2	2	2	2	8	2	2	7	8	2	2	2	2	8	8	8	8	2
		Σ	ıı	11	_	=	-	=	=	-	<u>-</u>	=	-	۲	_	=	<u>-</u>	÷	=	=	=	=	=	=	=	=	=
55		8	541	542	543	544	545	246	547	548	249	250	551	552	553	554	555	929	222	258	529	260	561	295	563	564	565

5		B-R8	I	I	I	I	I	I	I	I	Ξ	I.	I.	I	I	I	I	I	I	I	Ι	I	I	r	I	æ]	I
10		B-R7	CH3	I	Ŧ	I	I	I I	I.	I	Ξ	C4H9	C4H9	C4H9	C4H9	C4H9	C4H9	C4H9	С4Н9	C4H9	C4H9	C4H9	С4Н9	C4H9	C4H9	C4H9	C4H9
,,,		B-R8	I	I	I	I,	I	I	T.	I	I	I	Ξ	I	H	Н	I	I	н	Ι	π	Ι	I	I	I	I	I
15		B-R5	Н	I	I	Ξ	I	I	I	I	н	I	I	I	H	H	н	I	I	н	H	н	н	I	I	I	I
		A-R4	н	CF3	CF3	CF3	SF3	CF3	CF3	CF3	I	I	I	I	Ξ,	Ŀ	I	ш	I	F	, 4	H	CF3	CF3	I	I	I.
20		A-R3	F	I	ェ	I	I	I.	I	I	I	I	I	I,	ш,	I	щ	I	щ	I,	u.	I	I	I.	F	F	I
25		A-R2	CF3	CF3	CF3	CF3	CF3	CF3	CF3	CF3	CF3	I	I	ш	I	I	I	ட	ъ	I	ш.	CF3	I	CF3	CF3	CF3	CF3
23	tinued)	A-R1	I	I	I	I	I	I	I	I	I	I	ட	I	H	I	щ	I	I	L	ıL	I	I	I	I	ш	ш
30	Table 1-10 (continued)	ပ	СНЗ	СНЗ	СНЗ	СНЗ	CF3	CF3	СНЗ	C4H9	СНЗ		•	٠	-	•	•	•					•				
	Table 1	ſ	C2H5	Ξ	СНЗ	F	СНЗ	T.	CF3	F	C2H5		•	·	•	•	•	•	•	•					,	-	
35		ш	CH3	СНЗ	СНЗ	СНЗ	CF3	CF3	СНЗ	C4H9	СНЗ		•			•	•	•	•				,				
40		B'orB						•				Ы	P1	Ы	P1	Ы	Ρ.	P	F	Ε.	E	٦	표	Ē	Ξ.	표.	P1
		В	P.	P1	P1	P.	Ы	E	F	F	Ы	ы	P1	P1	P1	P1	F	<u>.</u>	F	2	E	٦	2	٦	7	2	P .
45		٧	F.	옵	Ph	ď	ዊ	문	F.	P.	-F	Æ	F.	Ph	Ph	P.	윤	Æ	돈	문	둡	둡	둡	듄	Æ	Æ	Æ
		c	-	-	1	-	-	-	-	-	-	-	1	1	,	1	1	, ,	-	-	-	-	-	-	-	-	-
50		٤	2	2	7	7	ď	7	2	7	7	2	2	2	2	. 2	7	7	7	2	2	7	7	2	8	2	2
		Σ	<u>-</u>	<u>-</u>	=	느	<u>-</u>	=	-	=	<u>-</u>	ı	ı	<u>-</u>	ı.	, 	1	ıı	=	=	=	=	=	=	=	=	= "
55		å	266	267	268	269	920	57.	572	573	574	575	576	277	578	279	280	581	285	583	284	585	586	287	288	289	290

5		B-R8	H	I	I	I	I	I	I	I	I	I
10		B-R7	C4H9	C4H9	C4H9	C4H9	C4H9	C4H9	C4H9	C4H9	C4H9	I
70		B-R8	I	I	I	I	I	I	I	I	I	Ι
15		B-R5	I	I	Ξ	Ι	Ξ	I	I	I	I	I
		A-R4	CF3	CF3	CF3	Ξ	I	I	I	I	ı	I
20		A-R3	F	I	u.	I	CH3	I	C2H5	I	C4H9	Ξ
25		A-R2	I	I	I	CH3	I	C2H5	I	C4H9	I	H
25	tinued)	A-R1	I	ш	ш	I	Ξ	I	I	I	I	ц
30	Table 1-10 (continued)	၁			:		,					
	Table 1	ſ		•		,	,		•		•	
35		Е				•		•				
40		B'orB	집	P	٣	P.	Ы	P1	Ы	P1	P	P1
		8	£	P	P1	Ы	Ы	ы	P1	P1	P1	P1
45		∢	Æ	Ph								
		c	-	٠	١	ı	+	1	1	1	1	١
50		ε	2	2	2	2	2	2	2	2	2	2
		Σ	ı	Ir	Ir	ır	lr	ır	Ir	Ir	Ir	1
55		o _N	591	592	593	594	595	969	597	869	669	009

Table 1-11

		· · · · ·		· · · ·	1		· · · · · · · ·		able 1-11						····
	No	М	m	n	A	8	B'orB"	A-R1	A-R2	A-R3	A-R4	B-R5	B-R6	B-R7	B-R8
5	601	lr	2	1	Ph	P1	P1	Н	F	Н	н	Н	Н	Н	Н
	602	lr	2	1	Ph	P1	P1	Н	Н	F	Н	H	Н	Н	н
	603	ir	2	1	Ph	P1	P1	Н	Н	Н	F	H	Н	н	Н
40	604	Ir	2	1	Ph	P1	P1	Н	F	Н	F	Н	Н	Н	Н
10	605	Ir	2	1	Ph	P1	P1	Н	F	F	Н	Н.	Н	нн	
	606	Ir	2	1	Ph	P1	P1	F	н `	Н	F	Н	Н	н	н
	607	lr	2	1	Ph	P1	P1	F	F	F	F	Н	Н	Н	Н
15	608	lr	2	1	Ph	P1	P1	Н	CF3	H	Н	Н	Н	Н	Н
	609	lr	2	1	Ph	P1	P1	Ι	Τ	Ι	CF3	Ξ	Н	H	Н
	610	lr	2	1	Ph	P1	P1	H	CF3	Ξ	CF3	Ξ	Н	Н	Н
	611	Ir	2	1	₽h	P1	P1	I	CF3	F	Н	Н	Н	Н	Н
20	612	lr	2	1	Ph	P1	P1	F	CF3	F	Н	Н	Н	Н	Н
	613	lr	2	1	Ph	P1	P1	F	CF3	Н	Н	_ Н	Н	Н	Н
	614	lr	2	1	Ph	P1	P1	H	Н	F	CF3	Н	н	Н	Н
25	615	lr	2	1	Ph	P1	P1	F	Н	Н	CF3	Н	Н	Н	н
	616	lr	2	1	Ph	P1	P1	F	Н	F	CF3	Н	н	Н	Н
	617	Ir	2	1	Ph	P1	P1	Н	СНЗ	Н	Н	Н	Н	Н	Н
20	616	lr	2	1	Ph	P1	P1	н	Н	СНЗ	Н	Н	Н	Н	Н
30	619	Ir	2	1	Ph	P1	P1	Н	C2H5	Н	Н	Н	Н	Н	Н
	620	lr	2	1	Ph	P1	P1	Н	Н	C2H5	Н	Н	Н	Н	Н
	621	lr	2	1	Ph	P1	P1	Н	C4H9	Н	Н	Н	Н	н	Н
35	622	lr	2	1	Ph	Pt	P1	Н	Н	C4H9	Н	н	Н	Н	н
,	623	lr	2	1	Ph	P1	P1	Н	Н	Н	Н	Н	Н	н	СНЗ
	624	lr	2	1	Ph	P1	P1	F	Н	Н	Н	Н	Н	Н	∗СН3
40	625	lr	2	1	Ph	P1	P1	Н	F	Н	Н	н	н	Н	- СН3
40	626	lr	2	1	Ph	P1	P1	Н	Н	°F	Н	н	Н	Н	-CH3
	627	ir	2	1	Ph	P1	P1	Н	H	Н	F	Н	Н	Н	СНЗ
	628	İr	2	1	Ph	P1	P1	F	н	F	Н	н	Н	Н	СНЗ
45	629	ir	2	1	Ph	P1	P1	Н	F	Н	F	Н	Н	Н	СНЗ
	630	İr	2	1	Ph	P1	P1	Н	F.	F	Н	Н	Н	Н	СНЗ
	631	lr	2	1	Ph	P1	P1	F	Н	Н	F	Н	Н	Н	СМЗ
50	632	lr	2	1	Ph	P1	P1	F	F	F	F	Н	Н	Н	СНЗ
50	633	lr	2	1	Ph	P1	P1	Н	CF3	Н	Н	Н	Н	Н	СНЗ
	634	lr	2	1	Ph	P1	P1	Н	H'·	Н	·CF3	н	Н	Н	СНЗ
l	635	lr :	2	1	Ph	P1	P1	Н	CF3	Н	CF3	Н	Н	Н	СНЗ
55	636	lr	2	1	Ph	P1	P1	Н	CF3	F	Н	Н	Н	Н	СНЗ
	637	lr	2	1	Ph	P1	P1	F	CF3	F	Н	Н	Н	Н	СНЗ
-															

Table 1-11 (continued)

	No	М	ш	n	Α	8	B'orB"	A D4	A D0	4 00	4 04	D DE	0.00	D D7	
	600					0	BUID	A-R1	A-R2	A-R3	A-R4	B-R5	B-R6	B-R7	B-R8
	638	lr	2	1	Ph	P1	P1	F	CF3	Н	Н	H	Η.	Ξ	СНЗ
5	639	lr	2	1	Ph	P1	P1	Н	H	F	CF3	H	Н	Η.	СНЗ
	640	lr	2	1	Ph	P1	P1	F	Н	Н	CF3	Н	Н	I	СНЗ
ſ	641	Ir	2	1	Ph	P1	P1	F	I	F	CF3	H	Н	Н	СНЗ
10	642	lr	2	1	Ph	P1	P1	Н	СНЗ	Н	Н	Н	н	н	СНЗ
Γ	643	lr	2	1	Ph	P1	P1	Н	Н	СНЗ	Н	Н	Н	Н	СНЗ
Γ	644	lr	2	1	Ph	P1	P1	Н	C2H5	Н	Н	Н	Н	н	-СН3
	645	lr	2	1	Ph	P1	P1	Н	Н	C2HS	Н	Н	н	Н	СНЗ
15	646	lr	2	1	Ph	P1	P1	н	C4H9	Н	Н	Н	Н	Н	СНЗ
	647	lr	2	1	Ph	P1	P1	н	н	C4H9	¹H	н	Н	Н	СНЗ
Ī	648	lr	3	0	Ph	P2	-	Н	н	СНЗ	Н	Н	Н	Н	•
20	649	lr	3	0	Ph	P2	-	Н	Н	C4H9	Н	н	Н	Н	•
ſ	650	lr	3	0	Ph	P2	-	F	Н	F	Н	Н	Н	Н	-
ſ	651	lr	3	0	Ph	P2	-	Н	Н	F	H	Н	Н	Н	•
[652	lr	3	0	Ph	P2	-	H	CF3	Н	Н	Н	Н	H	-
25	653	lr	3	0	Ph	P2	-	Н	Н	Η	Н	Н	Н	Н	•
	654	lr	3	0	Ph	P2	-	Н	Ξ	Н	Ή	Н	H	Η	-
	655	lr	3	0	Ph	P2	-	Н	Н	Η	Ħ	Н	Н	Н	
30	656	lr	3	0	Ph	P2	-	Н	Η	H	Η	Н	Н	•СН3	-
	657	lr	3	0	Ph	P2	-	Н	H	Н	Н	Н	СНЗ	Н	•
	658	lr	3	0	Ph	РЗ	-	Н	Н	СНЗ	Н	Н	Н	Н	
35	659	lr	3	0	Ph	РЗ	-	Н	Н	C4H9	Н	Н	Н	Н	
<i>"</i> [660	lr	3	0	Ph	РЗ	-	F	Η	F	Н	Н	Н	Н	•

Table 1-12

					2	IADIE 1-1								
B-R8	B-R7	B-R6	B-R5	A-R4	A-R3	A-R2	A-R1	В	Α	n	E	М	No	40
] H	н	н	Н	Ξ	L.	Τ	Η	РЗ	Ph	0	က	lr	661	
-	Н	н	н	Н	н	CF3	Н	P3	Ph	. 0	3	lr	662	
-	Н	Н	н	Н	Н	Н	Н	P3	Ph	0	3	lr	663	45
-	Н	Н	Н	Н	Н	н	Н	Р3	Ph	0	3	lr	664	
-	Н	Н	Н	H .	Н	Н	Н	Р3	Ph	0	3	ir	665	
] -	н	Н	СНЗ	Н	н.	Н	Н	Р3	Ph	0	3	lr	666	
-	Н	СНЗ	Н	Н	Н	Н	Н	Р3	Ph	0	3	lr	667	50
-	Н	н	Н	Н	СНЗ	Н	Н	P4	Ph	0	3	lr	668	
-	Н	н	Н	Н	C4H9	Н	Н	P4	Ph	0	3	lr	669	
-	н	н	Н	Н	F	Н	F	P4	Ph	0	3	Ir	670	55
-	н	Н	Н	Н	F	н	Н	P4	Ph	0	3	lr	671	
-	н	н	Н	Н	н	CF3	Н	P4	Ph	0	3	lr	672	
<u> </u>	Н	Н	Н	Н	F	н	Н	P4	Ph	0	3	lr	671	55

Table 1-12 (continued)

	No	М	m	n	Α	В	A-R1	A-R2	A-R3	A-R4	B-R5	B-R6	B-R7	B-R8
	673	ir	3	0	Ph	P4	Н	H	Н	Н	H			
5	674	lr	3	0	Ph	P4		Н		<u> </u>		Н	Н	-
	675		3	0	Ph		Н	ļ	Н	Н	Н	Н	Н	-
		lr	<u> </u>			P4	Н	н	Н	H.	Н	Н	H '	-
	676	lr	3	0	Ph	P4	Н	Н	Н	Н	СНЗ	Н	Н	-
10	677	lr	3	0	Ph	P4	Н	Н	Н	Н	Н	СНЗ	Н	
	678	lr -	3	0	Ph	P5	Н	СНЗ	Н	Н	Ŧ	СНЗ	Н	-
	679	lr_	3	0	Ph	P5	Н	Н	C4H9	Н	Н	Н	Н	-
15	680	lr	3	0	Ph	P5	Н	Н	F	Н	Н	Н	Н	-
13	681	Ir	3	0	Ph	P5	Н	CF3	Н	Н	Н	Н	Н	-
	682	lr_	3	0	Ph	P5	Η	Н	Н	Н	Ξ	∙СНЗ	Н	-
	683	lr	3	0	Ph	P6	Н	Н	СНЗ	Н	Н	Н	Н	Н
20	684	lr	3	0	Ph	P6	Н	Н	C4H9	Н	н	Н	Н	Н
	685	lr	3	0	Ph	P6	Н	Н	F	Н	Н	Н	Н	Н
	686	lr	3	0	Ph	P6	Н	CF3	Н	Н	H	Н	Н	Н
	687	lr	3	0	Ph	P6	Н	Н	Н	Н	Ι	СНЗ	Н	н
25	688	lr	3	0	Ph	P7	Н	Н	СНЗ	Н	Н	Н	Н	Н
	689	lr	3	0	Ph	P7	Н	Н	C 4H9	Н	Н	Н	Н	Н
	690	lr	3	0	Ph	P7	Н	Н	F	H	Н	Н	Н	Н
30	691	İr	3	0	Ph	P7	Н	CF3	Н	Н	Н	Н	Н	н
	692	lr	3	0	Ph	P7	Н	Н	Н	Н	Н	СНЗ	Н	Н
	693	lr	3	0	Ph	P8	Н	Н	СНЗ	Н	Н	Н	Н	Н
	694	lr	3	0	Ph.	P8	Н	н	C4H9	Н	Н	Н	Н	н
35	695	lr	3	0	Ph	P8	н	Н	F	Н	Н	Н	Н	Н
	696	lr	3	0	Ph	P9	Н	Н	н	Н	Н	Н	СНЗ	н
	697	lr	3	0	Ph	P8	Н	Н	Н	Н	Н	СНЗ	Н	Н
40	698	lr	3	0	Ph	P9	Н	Н	СНЗ	Н	Н	Н	Н	н
	699	lr	3	0	Ph	P9	Н	Н	C4H9	Н	Н	Н	Н	Н
	700	lr	3	0	Ph	P9	н	н	F	Н	н	н	Н	Н
,,	701	ir	3	0	Ph	P9	н	Н	Н	Н	Н	Н	СНЗ	Н
45	702	lr	3	0	Ph	P9	Н	Н	н	н	Н	СНЗ	н	н

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5	8RJ	=	Ξ		1	Ξ				-	Ŧ	Ŧ	=	=	1		F	ŀ	·	-		ŀ	•		ŀ	ŀ	-	·	\cdot		Ţ		F	3	=	
	2	=	F		†	<u> </u>		1			Ļ	Ŧ		¥ ¥	•	ļ		ŀ	·	•	. .	ŀ	·	1	ŀ	ŀ	ŀ	·	-	·			F	F	=	
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10	9	1			<u></u>	+	<u>_</u>	<u>.</u>	┇	+	+	Ī	=	-	+	╁	+		Ė	=	+	+	Ŧ	+	: -	+	×	, X	×	<u>-</u>	}	†	=			į
				×	<u>.</u>	+	Ц	_	2	+	+	-	Ŧ		<u> </u>	+	+	+	¥	z)	 	+	1	3	,		-	Ŧ	H	3	-	3	‡		3	
	9	+	╀	Н	+	+	Ŧ	Н	H	+	╁	╀	Н	H	4	-	+	-	Н	Н	+	+	Н	+	+			Ī	Н	Н	+	+	‡	_ 		
15	.)—	+	╀	Н	H	+	Н	Н	H	+	╁	╁	Н	Н	Н	H	+	+	H	Н	+	+	┞	Н	+	+		╀	Н	Н	+	┪	#	+	Н	
	2	+	╀	Ĭ	H	+	H	H	Н	+	╀	╀	Н	Н	H	=	+	╀	╀	Н	Н	╁	╁	Н	+	+	+	╀	<u>.</u>	-	Н	4	+	╀	H	
•		Н	Ϊ.	Ľ	ļ	+		Ħ		1	+	╀	7	Ĭ	H	ļ	7	<u> </u>	Ľ	H	Н	4	╀	Ľ	4	4	+	╀	┞	H	Н	Н	+	+	Ή	
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	1-17	Ε	~	•	2	2	2	7	2	2	ŀ	ı	•	ŀ	ŀ	Ŀ	٠	ŀ	8	2	2	7	2
50	ble l	3	4	٤	ď	£	٤	٧	4	¥	K	ď	٤	٤	٤	8	8	æ	М	2	2	2	2
	2	ž	힐	2	3	3	2	8	5	8	8	٤	=	2	F	Z	2	2	2	9	2	8	5

55 [0061] Hereinbelow, the present invention will be described more specifically based on Examples. [0062] Iridium metal coordination compounds used in Examples were synthesized along synthesis paths shown below. (Analogous reactions are described in Inorg. Chem. 1994, 33, p. 545).

<< Synthesis of iridium metal coordination compounds>>

[0063] A process scheme for synthesizing iridium complexes used in the present invention is shown below.

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(Example 1) Synthesis of Example Compound No. 729

[0064]

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$$S = B(OH)_2 + Br - CF_3$$

$$A$$

$$F_3C$$

$$F_3C$$

$$F_3C$$

$$F_3C$$

$$F_3C$$

$$F_3C$$

$$F_3C$$

[0065] In a 100 ml-three-necked flask, 3.18 g (24.9 mmol) of thienylboronic acid, 5.65 g-(25.0 mmol) of 1-bromo-4-trifluoromethylpyridine, 25 m of toluene, 12.5 ml of ethanol and 25 ml of 2M-sodium-carbonate aqueous solution, were placed and stirred at room temperature under a nitrogen stream, followed by addition of 0.98 g-(0.85 mmol) of tetrakis(triphenylphosphine)palladium (0). Thereafter, the system was refluxed under stirring and nitrogen stream for 8 hours. After completion of the reaction, the reaction product was cooled and extracted by adding cold water and toluene. The organic layer was washed with saline water and dried on magnesium sulfate, followed by removal of the solvent under a reduced pressure to provide dry solid. The residue was purified by silica gel-column chromatography (eluent: chloroform/methanol = 10/1) to obtain 4.20 g (yield = 74 %) of Compound A.

[0066] In a 100 ml-four-necked flask, 50 ml of glycerol was placed and heated for 2 hours at 130 - 140 °C under stirring and bubbling with nitrogen. The glycerol was cooled to room temperature and poured into 300 ml of 1N-hydrochloric acid, and the precipitate was filtered out and washed with water. The precipitate was then purified by silica gel chromatography with chloroform as eluent, to obtain 0.33 g (yield: 38 %) of red powdery Example Compound No. 729.

[0067] A toluene solution of the compound exhibited a luminescence spectrum showing \(\text{\text{max}} = 563 \) nm. The compound was subjected to MALDI-TOF (matrix-assisted laser desorption ionization time-of-flight mass spectroscopy) by using an apparatus ("REFLEX-III", made by Bruker Co.). In the method, an ion obtained by removing one electron from a sample substance is subjected to measurement of a mass thereof, so that the measured mass is denoted by M+, and the method is frequently used for identification of a substance. The measured M+ value was 877.0 from which the objective product was confirmed.

[0068] For confirmation of phosphorescence-type luminescence, the Example Compound was dissolved in chloroform, and the solution was separately aerated with oxygen or nitrogen, each followed by photoirradiation for comparison of photoluminescence. As a result, substantially no luminescence attributable to the iridium complex was recognized with respect to the oxygen-aerated solution, whereas photoluminescence was confirmed with respect to the nitrogen-aerated solution. From these results, the compound of the present invention was confirmed to be a phosphorescent compound. For reference, in the case of a fluorescent material, luminescence attributable to the compound does not disappear even in an oxygen-aerated solution.

[0069] Further, in contrast with a fluorescent material generally showing a luminescence life of several nsec to several tens of nsec, the compounds of the present invention including those obtained in the following Examples, all exhibited a phosporescence life of 100 nsec or longer.

(Example 2)

20 [0070] Example Compound No. 310 was synthesized through a similar process as in Example 1.

Luminescence of toluene solution: λmax = 489 nm

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MALDI-TOF MS: M^+ = 859.1

(Example 3)

30 [0071] Example Compound No. 238 was synthesized through a similar process as in Example 1.

Luminescence of toluene solution: $\lambda max = 515 \text{ nm}$

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MALDI-TOF MS: $M^+ = 709.1$

(Example 4)

40 [0072] Example Compound No. 242 was synthesized through a similar process as in Example 1.

Luminescence of toluene solution: $\lambda max = 471 \text{ nm}$

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MALDI-TOF MS: $M^+ = 763.1$

(Example 5)

[0073] Example Compound No. 384 was synthesized through a similar process as in Example 1.

Luminescence of toluene solution: λmax = 466 nm

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MALDI-TOF MS: $M^+ = 913.1$

(Example 6)

[0074] Example Compound No. 777 was synthesized through a similar process as in Example 1.

Luminescence of toluene solution: max = 696 nm

MALDI-TOF MS: M+ = 1.231.1

(Example 7)

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[0075] Example Compound No. 472 was synthesized.

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[0076] In a 100 ml-two-necked flask, 60 ml of ethoxyethanol and 20 ml of H₂O were placed and stirred for 1 hour under bubbling with nitrogen. Then, .51 g (4.4 mmol) of Compound C and 0.71 g (2.0 mmol) of indium (III) trichloride hydrate were added, and the system was heated for 16 hours around 100 °C under stirring and nitrogen stream. The reaction product was cooled to room temperature and poured into 100 ml of water, followed by recovery by filtration and washing with water of the precipitate. The precipitate was then poured into 60 ml of ethanol and stirred for 1 hour, followed by filtering-out and washing with acetone, to obtain 0.95 g (yield: 89 %) of yellow powdery Compound D.

[0077] In a 100 ml-two-necked flask, 50 ml of ethoxyethanol was placed and stirred for 1 hour under bubbling with nitrogen. Then, 0.536 g (0.5 mmol) of Compound D, 0.17 g (1.4 mmol) of Compound E and 0.75 g of sodium carbonate Na₂CO₃ were added, and the system was heated for 16 hours around 100 °C under stirring and nitrogen stream. The reaction product was cooled to room temperature and poured into 100 ml of water, followed by filtering-out and washing with water of the precipitate. The precipitate was poured into 70 ml of ethanol, and after stirring for 1 hour, the precipitate was filtered out and dissolved in chloroform, followed by filtration. The resultant filtrate was condensed, and purified by silica gel column chromatography with chloroform as eluent to obtain 0.45 g (yield: 73 %) of yellow powdery Example Compound No. 472. A toluene solution of the compound exhibited a luminescence spectrum showing \(\lambda\text{max} = \frac{526}{26}\text{ nm}\). The compound exhibited M+ = 614.2 according to MALDI-TOF MS and was confirmed to be the objective product.

(Example 8)

[0078] In this Example, a device (effective display area = 3 mm²) having a device structure including 4 organic layers as shown in Figure 1(c) was prepared. An alkali-free glass sheet was used as a transparent substrate 15 and a 100 nm-thick indium tin oxide (ITO) film was formed by sputtering and patterned as a transparent electrode 14. Further, α-NPD represented by the above-mentioned structural formula was vacuum-deposited in a layer thickness of 40 nm thereon as a hole-transporting layer 13. Then, as an organic luminescence layer 12, the above-mentioned CBP as a host material and Example Compound No. 729 (metal coordination compound) in an amount of providing 8 wt. % were co-vacuum deposited in a layer thickness of 30 nm. Further, as an exciton diffusion-prevention layer 17, BCP was vacuum-deposited in a thickness of 10 nm. Then, as an electron-transporting layer 16, the above-mentioned Alq3 was subjected to resistance heating vacuum deposition at a vacuum of 10-4 Pa to form an organic film in a thickness of 30 nm. [0079] On the above, as a lower layer of a metal electrode layer 11, an AlLi alloy film was disposed in a thickness of 15 nm, and a 100 nm-thick Al film was vacuum-deposited thereon to form a patterned metal electrode 11 disposed opposite to the transparent electrode 14 and having an electrode area of 3 mm².

[0080] The performances of the thus-obtained EL device were measured by using a micro-current meter ("4140B", made by Hewlett-Packard Corp.) for a current-voltage characteristic and "BM7" (made by Topcon K.K.) for an emission luminance.

(Example 9)

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[0081] A device was prepared in the same manner as in Example 8 except using a metal-coordination-compound (Example Compound No. 729) was used in a weight ratio of 7 wt. %.

(Comparative Example 1)

[0082] A device was prepared in the same manner as in Example 8 except for using a metal coordination compound (729R) shown in Table 2 (wherein a substituted compound of the present invention to be compared therewith is shown in parallel) in a weight ratio of 8 wt. %.

Table 2

No	М	N	m	Α	В	A-R1	A-R2	A-R3	A-R4	B-R5	B-R6	B-R7	B-R8
729R	lr	3	0	Tn1	P1	Н	Н	•	•	Н	Н	Н	Н
729	lr	3	0	Tn1	P1	Н	Н	-	-	Н	Н	℃F ₃	Н

(Comparative Example 2)

[0083] A device was prepared in the same manner as in Example 8 except for using the metal-coordination-compound (729R) shown in Table 2 in a weight ratio of 3 wt. %.

(Comparative Example 3)

[0084] A device was prepared in the same manner as in Example 8 except for using the metal coordination compound (729R) shown in Table 2 in a weight ratio of 1 wt.%.

[0085] Each device was supplied with an electric field of 12 volts/100 nm with the ITO side as the anode and the Al side as the cathode to measure a luminance.

[0086] In order to remove factors for device deterioration due to oxygen or water, the above-measurement was performed in a dry nitrogen flow after taking the device out of the vacuum chamber.

[0087] The results of devices using the respective compounds are shown in Table 3. As is understood from the results shown in Table 3, the maximum luminance concentration of Comparative Compound 729R was clearly between 1 % and 8 %, whereas Example Compound No. 729 provided with a substituent exhibited a higher luminance at 8 % than at 7 % and could exhibit a substantially higher luminance at 8 % than 729R having no substituent.

Table 3

<Luminance comparison> Example Compound No. Concentration (wt.%) Luminance (cd/m2) 8 729 8 4500 9 729 7 4250 Comp. 1 729R 8 1620 Comp. 2 729R 3 4000 Comp. 3 729R 1 1290

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(Example 10)

[0088] A device was prepared in the same manner as in Example 8 except for using a metal-coordination compound (310) synthesized in Example 2 in a weight ratio of 3 wt. %.

(Example 11)

[0089] A device was prepared in the same manner as in Example 8 except for using a metal coordination compound (310) synthesized in Example 2 in a weight ratio of 6 wt. %.

(Example 12)

[0090] A device was prepared in the same manner as in Example 8 except for using a metal coordination-compound (310) synthesized in Example 2 in a weight ratio of 8 wt. %.

(Example 13)

[0091] A device was prepared in the same manner as in Example 8 except for using a metal coordination compound (238) synthesized in Example 3 in a weight ratio of 3 wt. %.

(Example 14)

[0092] A device was prepared in the same manner as in Example 8 except for using a metal coordination-compound (238) synthesized in Example 3 in a weight ratio of 6 wt. %.

(Example 15)

[0093] A device was prepared in the same manner as in Example 8 except for using a metal coordination compound (238) synthesized in Example 3 in a weight ratio of 8 wt. %.

(Example 15A)

[0094] A device was prepared in the same manner as in Example 8 except for using a metal coordination compound (238) synthesized in Example 3 in a weight ratio of 11 wt. %.

(Example 15B)

[0095] A device was prepared in the same manner as in Example 8 except for using a metal coordination compound (238) synthesized in Example 3 in a weight ratio of 13 wt. %.

(Example 16)

[0096] A device was prepared in the same manner as in Example 8 except for using a metal coordination compound (242) synthesized in Example 4 in a weight ratio of 3 wt. %.

(Example 17)

[0097] A device was prepared in the same manner as in Example 8 except for using a metal coordination compound (242) synthesized in Example 4 in a weight ratio of 6 wt. %.

(Example 18)

[0098] A device was prepared in the same manner as in Example 8 except for using a metal coordination-compound (242) synthesized in Example 4 in a weight ratio of 8 wt. %.

(Example 19)

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[0099] A device was prepared in the same manner as in Example 8 except for using a metal coordination-compound (384) synthesized in Example 5 in a weight ratio of 3 wt. %.

(Example 20)

[0100] A device was prepared in the same manner as in Example 8 except for using a metal coordination-compound (384) synthesized in Example 5 in a weight ratio of 6 wt. %.

(Example 21)

[0101] A device was prepared in the same manner as in Example 8 except for using a metal coordination compound (384) synthesized in Example 5 in a weight ratio of 8 wt. %.

(Comparative Example 4)

[0102] A device was prepared in the same manner as in Example 8 except for using a metal coordination-compound (1R) shown in Table 4 (wherein structures of the corresponding Example Compound Nos. 310, 238, 242 and 384 are shown in parallel) in a weight ratio of 3 wt. %.

Table 4

No	М	m	n	Α	В	A-R1	A-R2	A-R3	A-R4	B-R5	B-R6	3-R7	B-R8
1R	lr	3	0	Ph	.P1	Н	Н	Н	Н	Н	Н	Н	Н
310	lr	3	0	Ph	P1	Н	CF ₃	Н	Н	Н	Н	Н	Н
238	lr	3	0	Ph	P1	Н	F	Н	Н	Н	Н	Н	н
242	lr	3	0	Ph	P1	F	Н	F	Н	Н	Н	н	Н
384	lr	3	0	Ph	P1	Н	CF ₃	F	Н	Н	Н	н	Н

(Comparative Example 5)

[0103] A device was prepared in the same manner as in Example 8 except for using the metal coordination compound (1R) shown in Table 4 in a weight ratio of 6 wt. %.

(Comparative Example 6)

[0104] A device was prepared in the same manner as in Example 8 except for using the metal coordination compound (1R) shown in Table 4 in a weight ratio of 8 wt. %.

[0105] Each device of Examples 10 - 12 and Comparative-Examples 4 - 6 was supplied with an electric field of 12 volts/100 nm with the ITO side as the anode and the Al side as the cathode to measure a current efficiency.

[0106] The results of devices using the respective compounds are shown in Table 5. As is understood from the results shown in Table 5, the concentration giving a maximum current efficiency of Comparative Compound 1R was clearly between 3 % and 8 %, whereas Example Compound No. 310 provided with a substituent exhibited an increase in current efficiency even at 8 %.

Table 5

	<compari< th=""><th>son of current efficiency:</th><th>></th></compari<>	son of current efficiency:	>
Example	Compound No.	Concentration (wt.%)	Current Eff. (cd/A)
10	310	3	2
11	310	6	2.4
12	310	8	2.7
Comp. 4	1R	3	15
Comp. 5	1R	6	19
Comp. 6	1R	8	17

[0107] Each device of Examples 13 - 15 and Comparative Examples 4 - 6 was supplied with a voltage of 12 volts/ 100 nm with the ITO side as the anode and the Al side as the cathode to measure an (electric) power efficiency. [0108] The results of the devices using the respective compounds are shown in Table 6. As is understood from the results shown in Table 6, the concentration giving a maximum power efficiency of the Comparative Compound 1R was between 3 % and 8 %, whereas Example Compound (238) provided with a substituent showed an increase in maximum efficiency even at a concentration of 8 %.

Table 6

	<comp< th=""><th>parison of power efficient</th><th>cy></th></comp<>	parison of power efficient	cy>
Example	Compound No.	Concentration (wt.%)	Power efficiency (lm/W)
13	- 238	3	5.4
14	238	6	6
15	238	8	6.2
15A	238	11	6.5
15B	238	13	6.3
Comp. 4	1R	3	5.7
Comp. 5	1R	6	6.2
Comp. 6	1R	8	6

[0109] Each device of Examples 16 - 18 and Comparative Examples 4 - 6 was supplied with a voltage of 12 volts/
100 nm with the ITO side as the anode and the Al side as the cathode to measure an external quantum efficiency, which was evaluated in terms of a ratio of luminance (lm)/current (mA) based on a value of current passing through the device by using a micro-current passing through the device by using a micro-current meter ("4140B", made by Hewlett-Packard Corp.) and a value of luminance measured by "BM7" (made by Topcon K.K.).

[0110] The results of the devices using the respective compounds are shown in Table 7. As is understood from the results shown in Table 7, the concentration giving a maximum external quantum efficiency of the Comparative Compound 1R was between 3 % and 8 %, whereas Example Compound (242) provided with a substituent showed an increase in maximum efficiency even at a concentration of 8 %.

Table 7

	<comparison< th=""><th>n of external quantum ef</th><th>ficiency></th></comparison<>	n of external quantum ef	ficiency>
Example	Compound No.	Concentration (wt.%)	Ext. quantum efficiency
16	242	3	3
17	242	6	4
18	242	8	4.2
Comp. 4	1R	3	7
Comp. 5	1R	6	8
Comp. 6	· 1R	-8	7.6

[0111] Each device of Examples 19 - 21 and Comparative Examples 4 - 6 was supplied with a voltage of 12 volts/ 100 nm with the ITO side as the anode and the Al side as the cathode to measure an (electric) power efficiency.

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[0112] The results of the devices using the respective compounds are shown in Table 8. As is understood from the results shown in Table 8, the concentration giving a maximum power efficiency of the Comparative Compound 1R was between 3 % and 8 %, whereas Example Compound (384) provided with a substituent showed an increase in maximum efficiency even at a concentration of 8 %.

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Table 8

	<comp< th=""><th>parison of power efficient</th><th>cy></th></comp<>	parison of power efficient	cy>
Example	Compound No.	Concentration (wt.%)	Power efficiency (lm/W)
19	384	3	2
20	384	6	-2.3
21	384	8	2.6
Comp. 4	1R	3	5.7
Comp. 5	1R	6	6.2
Comp. 6	1R	8	· 6

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(Example 22)

20 [0113] A device was prepared in the same manner as in Example 8 except for using a metal coordination-compound (777) synthesized in Example 6 in a weight ratio of 1 wt. %.

(Example 23)

25 [0114] A device was prepared in the same manner as in Example 8 except for using a metal-coordination-compound (777) synthesized in Example 6 in a weight ratio of 6 wt. %.

(Example 24)

[0115] A device was prepared in the same manner as in Example 8 except for using a metal-coordination-compound (777) synthesized in Example 6 in a weight ratio of 8 wt. %.

(Comparative Example 7)

35 [0116] A device was prepared in the same manner as in Example 8 except for using a metal coordination compound (777R) shown in Table 9 below in a weight ratio of 1 wt. %.

Table 9

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No	М	m	n	Α	В	A-R1	A-R2	A-R3	A-R4	3-R5	3-R6	B-R7	B-R8
777R	lr	3	0	Pe	P1	Н	Н	Н	Н	Н	Н	Н	Н
777	ir	3	0	Pe	P1	Н	Н	Н	н	н	Н	:CF ₃	Н

(Comparative Example 8)

[0117] A device was prepared in the same manner as in Example 8 except for using the metal coordination compound (777R) shown in Table 9 in a weight ratio of 6 wt. %.

(Comparative Example 9)

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[0118] A device was prepared in the same manner as in Example 8 except for using the metal coordination compound (777R) shown in Table 9 in a weight ratio of 8 wt. %.

[0119] Each device of Examples 22 - 25 and Comparative Examples 7 - 9 was supplied with a voltage of 42 volts/ 100 nm with the ITO side as the anode and the Al side as the cathode to measure an (electric) power efficiency.

[0120] The results of the devices using the respective compounds are shown in Table 10. As is understood from the results shown in Table 6, the concentration giving a maximum power efficiency of Comparative Compound 777R was between 1 % and 8 %, whereas Example Compound (777) provided with a substituent showed an increase in maximum

efficiency up to a concentration of 8 %.

Table 10

	<compariso< th=""><th>on of maximum power eff</th><th>iciency></th></compariso<>	on of maximum power eff	iciency>
Example	Compound No.	Concentration (wt.%)	Power efficiency (lm/W)
22	777	1	0.04
23	777	6	0.12
24	777	8	0.15
Comp. 7	777R	1	0.08
Comp. 8	777R	6	0.15
Comp. 9	.777R	8	0.13

15 (Example 25)

[0121] A device was prepared in the same manner as in Example 8 except for using a metal-coordination compound (472) synthesized in Example 7 in a weight ratio of 3 wt. %.

20 (Example 26)

[0122] A device was prepared in the same manner as in Example 1 except for using a metal coordination compound (472) synthesized in Example 7 in a weight ratio of 6 wt. %.

25 (Comparative Example 10)

[0123] A device was prepared in the same manner as in Example 1 except for using a metal coordination compound (472R) shown below in a weight ratio of 3 wt. %.

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$$\begin{bmatrix} \bigcirc \\ \bigcirc \\ \bigcirc \\ \end{bmatrix}_2 \text{ o-c} CH_3$$

$$CH_3$$

$$CH_3$$

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(Comparative Example 11)

[0124] A device was prepared in the same manner as in Example 1 except for using the above metal coordination compound (472R) in a weight ratio of 6 wt. %.

(Comparative Example 12)

[0125] A device was prepared in the same manner as in Example 1 except for using the above metal coordination compound (472R) in a weight ratio of 8 wt. %.

[0126] Each device of Examples 25 - 27 and Comparative Examples 10 - 12 was supplied with an electric field of 12 volts/100 nm with the ITO side as the anode and the Al side as the cathode to measure a power efficiency.

[0127] In order to remove factors for device deterioration due to oxygen or water, the above-measurement was performed in a dry nitrogen flow after taking the device out of the vacuum chamber.

[0128] The results of devices using the respective compounds are shown in Table 11. As is understood from the results shown in Table 11, the concentration giving a maximum power efficiency of Comparative Compound 1R was clearly between 3 % and 8 %, whereas Example Compound (384) provided with a substituent exhibited an increase in power efficiency even at a concentration of 8 %.

Table 11

<comparison efficiency="" maximum="" of="" power=""></comparison>					
Example No.		Concentration (wt.%)	Power efficiency (1m/W)		
25	472	3	5.6		
26	472	6	6.3		
27	472	8	6.5		
Comp. 10	472R	3	5:4		
Comp. 11	472R	6	6		
Comp. 12	472R	8	5.8		

(Example 28)

[0129]

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[0130] In a 200 ml-three-necked flask, 3.50 g (25.0 mmol) of 4-fluorophenylboronic acid, 3.95 g (25.0 mmol) of 1-bromopyridine, 25 ml of toluene, 12.5 ml of ethanol and 25 ml of 2M-sodium carbonate aqueous solution, were placed and stirred at room temperature under a nitrogen stream, followed by addition of 0.98 g (0.85 mmol) of tetrakis(triphenylphosphine)palladium (0). Thereafter, the system was refluxed under stirring and nitrogen stream for 8 hours. After completion of the reaction, the reaction product was cooled and extracted by adding cold water and toluene. The organic layer was washed with saline water and dried on magnesium sulfate, followed by removal of the solvent under a reduced pressure to provide dry solid. The residue was purified by silica gel column chromatography (eluent: chloroform/methanol = 10/1) to obtain 3.24 g (yield = 75 %) of Compound G.

[0131] In a 200 ml-three-necked, 0.881 g (2.5 mmol) of iridium (III) chloride trihydrate, 0.953 g (5.5 mmol), 75 ml of ethoxyethanol and 25 ml of water were placed and stirred for 30 min. at room temperature under nitrogen stream, followed by 24 hours of reflux under stirring. The reaction product was cooled to room temperature, and the precipitate was recovered by precipitation and washed successively with water, ethanol and acetone. After being dried at room temperature under a reduced pressure, 1.32 g (yield: 92 %) of yellow powdery Compound H was obtained.

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[0132] In a 200 ml-three-necked flask, 70 ml of ethoxyethanol, 0.80 g (0.7 mmol) of Compound H, 0.22 g (2.10 mmol) of acetylacetone and 1.04 g (9.91 mmol) of sodium carbonate, were placed and stirred for 1 hour at room temperature under a nitrogen stream, followed by 15 hours of reflux under stirring. The reaction product was cooled with ice, and the precipitate was filtered out and washed with water. The precipitate was purified by silica gel chromatography (eluent: chloroform/methanol = 30/1) to obtain 0.63 g (yield: 71 %) of yellow powdery Compound I (Example Compound No. 489). A toluene solution of the compound exhibited a luminescence spectrum showing λ max = 499 nm. Further, according to MALDI-TOF MS, M+ = 638.7 of the compound was confirmed.

[0133] In a 100 ml-three-necked flask, 0.21 g (1.2 mmol) of Compound G, 0.32 g (0.5 mmol) of Compound I and 25 ml of glycerol, were placed and stirred for 8 hours around 180 °C under a nitrogen stream. The reaction product was cooled to room temperature and poured into 170 ml of 1N-hydrochloric acid. The precipitate was filtered out and washed with water, followed by drying for 5 hours at 100 °C under a reduced pressure. The precipitate was purified by silica gel column chromatography with chloroform as the eluent to obtain 0.22 g (yield: 63 %) of yellow powdery Example Compound No. 239. A toluene solution of the compound exhibited a luminescence spectrum showing λ max = 490 nm, and M+ = 708.8 of the compound was confirmed by MALDI-TOF MS.

(Example 29)

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[0134] Example Compound No. 535 was synthesized through a similar process as in Example 7.

Luminescence of toluene solution: λmax = 525 nm

MALDI-TOF MS: M+ = 671.1

(Example 30)

[0135] Example Compound No. 243 was synthesized through a similar process as in Example 28.

55 Luminescence of toluene solution: λmax = 518 nm

MALDI-TOF MS: $M^+ = 762.7$

(Example 31) 5 [0136] Example Compound No. 511 was synthesized through a similar process as in Example 7. Luminescence of toluene solution: λmax = 514 nm 10 MALDI-TOF MS: M+ = 1628.1 (Example 32) 15 [0137] Example Compound No. 56 was synthesized through a similar process as in Example 28. Luminescence of toluene solution: $\lambda max = 505 \text{ nm}$ 20 MALDI-TOF MS: M^+ = 697.2 (Example 33) [0138] Example Compound No. 389 was synthesized through a similar process as in Example 1. Luminescence of toluene solution: λmax = 503 nm 30 (Example 34) [0139] Example Compound No. 390 was synthesized through a similar process as in Example 1. 35 Luminescence of toluene solution: λmax = 507 nm (Example 35) [0140] Example Compound No. 312 was synthesized through a similar process as in Example 1. [0141] The Luminescence of toluene solution exhibited two peaks at 458 nm and 488 nm. (Example 36) [0142] Example Compound No. 312 is synthesized through a similar process as in-Example 1. (Example 37) [0143] Example Compound No. 314 is synthesized through a similar process as in Example 1. 50 (Example 38) [0144] Example Compound No. 388 is synthesized through a similar process as in Example 1. (Example 39) [0145] Example Compound No. 392 is synthesized through a similar process as in Example 1.

(Example 40)

[0146] Example Compound Nos. 274, 346, 358, 393 and 396 can be synthesized through a similar process except for changing the starting material.

(Example 41)

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[0147] Hereinbelow, two examples of display apparatus are described. First, an example of preparation of a picture display apparatus having an XY-matrix structure is described with reference to Figure 2.

- [0148] On a glass substrate 21 measuring 150 mm-length, 150 mm-width and 1.1 mm-thickness, a ca. 100 nm-thick ITO film was formed by sputtering and patterned into 100 lines of 100 μm-wide transparent matrix electrodes (anode side) with a spacing of 40 μm as simple matrix electrodes. Then, a four-layered organic compound layer 23 was formed thereon including a luminescence layer 12 containing one of the compounds synthesized in Examples 1 7 as a guest compound.
- (0149) Then, 100 lines of 100 μm-wide metal electrodes 24 were formed with a spacing of 40 μm by mask vacuum deposition so as to be perpendicular to the transparent electrodes by vacuum deposition at a vacuum of 2x10-5 Torr. The metal electrodes were formed as a lamination of 10 nm-thick layer of Al/Li alloy (Li: 1.3 wt. %) and then 150 nm-thick layer of A1.
 - [0150] The thus-obtained 100x100-simpe matrix-type organic EL devices were subjected to a simple matrix drive in a glove box filled with nitrogen at voltages of 7 volts to 13 volts by using a scanning signal of 10 volts and data signals of ±3 volts as shown in Figure 3. As a result of an interlaced drive at a frame frequency of 30 Hz, luminescence pictures were confirmed for the respective devices.
 - [0151] As a picture display apparatus, the high-efficiency luminescence device of the present invention allows a light-weight flat panel display with economized energy consumption and high-recognizability. As a printer light source, the luminescence devices of the present invention may be arranged in a line and disposed in proximity to the photosensitive drum, to provide a line shutter wherein the respective devices are driven independently from each other to effect prescribed exposure on the photosensitive drum. On the other hand, the energy consumption economization effect is expected in application as an illumination device or a backlight for a liquid crystal display apparatus.
 - [0152] For another application to a picture display device, it is particularly advantageous to form an active matrix-type picture display device equipped with thin film transistors (TFTs) instead of the above-mentioned XY-matrix wiring. Hereinbelow, an active matrix-type picture display device according to the present invention will be described with reference to Figures 4 to 6.
 - [0153] Figure 4 is a schematic plan view of such a panel. Circumferentially outside the panel are disposed a drive circuit comprising a power supply source and a scanning signal driver, and a data signal driver as a display signal input means (called a picture data supply means, which are respectively connected to current supply lines, X-direction scanning lines called gate lines and Y-direction lines called data lines. The scanning signal driver sequentially selects the gate scanning lines, and in synchronism therewith, picture signals are supplied from the data signal driver. Display pixels are disposed at intersections of the gate scanning lines and the data lines.
 - [0154] Next, a pixel circuit operation is described with reference to an equivalent circuit. When a selection signal is applied to a gate selection line, TFT1 is turned on so that a data signal is supplied from a data signal line to a capacitor Cadd, thereby determining the gate potential of TFT2, whereby a current is supplied to an organic luminescence device (EL) disposed at each pixel through a current supply line depending on the gate potential of TFT2. The gate potential of TFT2 is held at Cadd during one frame period, so that the current continually flows from the current supply line to the EL device during the period. As a result, luminescence is retained during one frame period.
- 45 [0155] Figure 6 is a schematic view illustrating a sectional structure of a TFT used in this Example. On a glass substrate, a polysilicone p-Si layer is formed, and the channel, drain and source regions are doped with necessary impurities, respectively. Thereon, gate electrodes are formed via a gate insulating film, and drain electrodes and source electrodes connected to the drain regions and source regions, respectively, are formed. In this instance, the drain electrodes and transparent pixel electrodes (ITO) are connected through contact holes bored in an intervening insulating film.
 - [0156] The active device used in the present invention need not be particularly restricted, and can also be a single-crystal silicon TFT, an amorphous silicon a-Si TFT, etc.
 - [0157] On the pixel electrodes, plural layers or a single layer of organic luminescence layer may be disposed and metal electrodes as cathode are sequentially laminated to provide an active-type organic luminescence device.

[INDUSTRIAL APPLICABILITY]

[0158] As described above, a substituted metal coordination compound having a high phosphorescence efficiency

and a short phosphorescence life can be used in a luminescence layer at a high-concentration relative to the host material while preventing concentration extinction. As a result, according to the present invention, it is possible to obtain an excellent luminescence device showing high luminescence efficiency. The luminescence device of the present invention is also excellent as a display device.

Claims

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1. An organic luminescence device, comprising: a pair of electrodes each disposed on a substrate, and at least one luminescence layer comprising an organic compound disposed between the electrodes; wherein the luminescence layer comprises a non-luminescent first organic compound and a phosphorescent second organic compound represented by formula (1) shown below, and the second organic compound is present at a concentration of at least 8 wt. % in the luminescence layer:

$$ML_m L_n'$$
 (1),

wherein M is a metal atom of Ir, Pt, Rh or Pd; L and L' are mutually different bidentate ligands; m is 1, 2 or 3; n is 0, 1 or 2 with the proviso that m+n is 2 or 3; a partial structure ML_m is represented by formula (2) shown below and a partial structure ML_n is represented by formula (3), (4) or (5) shown below:

wherein N and C are nitrogen and carbon atoms, respectively; A and A' are respectively a cyclic group-capable of having a substituent and bonded to the metal atom M via the carbon atom; B, B' and B" are respectively a cyclic group represented by a formula of (6) - (14) shown below capable of having a substituent and connected to the metal atom M via the nitrogen atom:

{wherein the substituent is selected from a halogen atom, a cyano group, a nitro group, a trialkylsilyl group (of which the alkyl groups are independently a linear or branched alkyl group having 1 to 8 carbon atoms), a linear or branched alkyl group having 1 to 20 carbon atoms (of which the alkyl group can include one or non-neighboring two or more methylene groups that can be replaced with -O-, -S-, -CO-, -CO-O-, -O-CO-, -CH=CH- or -C=C- and the alkyl group can include a hydrogen atom that can be replaced with a fluorine atom), or an aromatic cyclic group capable of having a substituent (of which the substituent is selected from a halogen atom, a cyano group, a nitro group, a linear or branched alkyl group having 1 to 20 carbon atoms (of which the alkyl group can include one or non-neighboring two or more methylene groups that can be replaced with -O-, -S-, -CO-, -CO-O-, -O-CO-, -CH=CH-Or -C=C- and the alkyl group can include a hydrogen atom that can be replaced with a fluorine atom)};

A and B, and A' and B' are respectively bonded to each other via a covalent bond:

E and G are independently a linear or branched alkyl group having 1 to 20 carbon atoms (of which the alkyl group can include a hydrogen atom that can be optionally replaced with a fluorine atom), or an aromatic cyclic group capable of having a substituent (of which the substituent is selected from a halogen atom, a cyano group, a nitro group, a trialkylsilyl group (of which the alkyl groups are independently a linear or branched alkyl group having 1 to 8 carbon atoms), a linear or branched alkyl group having 1 to 20-carbon atoms (of which the alkyl group can include one or non-neighboring two or more methylene groups that can be replaced with -O-, -S-, -CO-, -O-CO-. -CH=CH- or -C=C- and the alkyl group can include a hydrogen atom that can be replaced with a fluorine atom).

J is a hydrogen atom, a halogen atom, a linear or branched alkyl group having 1 to 20 carbon atoms (of which the alkyl group can include a hydrogen atom that can be optionally replaced with a fluorine atom), or an aromatic cyclic group capable of having a substituent (of which the substituent is selected from a halogen atom, a cyano group, a nitro group, a trialkylsilyl group (of which the alkyl groups are independently a linear or branched alkyl group having 1 to 8 carbon atoms), or a linear or branched alkyl group having 1 to 20 carbon atoms (of which the alkyl group can include one or non-neighboring two or more methylene groups that can be replaced with -O-, -S-, -CO-O-, -O-CO-, -CH=CH- or -C=C- and the alkyl group can include a hydrogen atom that can be replaced with a fluorine atom));

wherein the compound represented by the formula (1) includes at least one cyclic group having a substituent.

- 2. An organic luminescence device according to Claim 1, comprising: a pair of electrodes each disposed on a substrate, and at least one luminescence layer comprising an organic compound disposed between the electrodes; wherein the luminescence layer comprises a non-luminescent first organic-compound and a phosphorescent second organic compound represented by the above-mentioned formula (1), and the second organic compound is present at a concentration in the luminescence layer that is higher than a concentration at which an cyclic group represented by the formula (1) but containing no substituent in any of the cyclic groups A and A' or the cyclic groups
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 B and B' exhibits a maximum luminescence characteristic.
 - 3. An organic luminescence device according to Claim 1, comprising: a pair of electrodes each disposed on a substrate, and at least one luminescence layer comprising an organic compound disposed between the electrodes; wherein the luminescence layer comprises a non-luminescent first organic compound and a phosphorescent second organic compound represented by the above-mentioned formula (1), and the second organic compound is present at a prescribed concentration of at least 8 wt. % in the luminescence layer providing a maximum luminescence characteristic.
- 4. An organic luminescence device according to Claim 1, wherein the partial structure ML'_n in the formula-(1) is represented by the formula (3).
 - An organic luminescence device according to Claim 1, wherein the partial structure ML'_n in the formula -(1) is represented by the formula (4).
- An organic luminescence device according to Claim 1, wherein the partial structure ML'_n in the formula (1) is represented by the formula (5).
 - 7. An organic luminescence device according to Claim 1, wherein n in the formula (1) is 0.
- An organic luminescence device according to Claim 1, wherein the substituent of the compound of the formula (1) is fluorine.
 - 9. An organic luminescence device according to Claim 1, wherein the substituent of the-compound of the formula (1)

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is a trifluoromethyl group.

- 10. An organic luminescence device according to Claim 1, wherein the substituent of the compound of the formula (1) is an alkyl group.
- 11. An organic luminescence device according to Claim 2, wherein said maximum luminescence characteristic is a maximum luminescence luminance.
- 12. An organic luminescence device according to Claim 2, wherein said maximum luminescence characteristic is a maximum current.
 - 13. An organic luminescence device according to Claim 2, wherein said maximum luminescence characteristic is an external luminescence efficiency.
- 14. An organic luminescence device according to Claim 2, wherein said maximum luminescence characteristic is a ratio of luminescence flux/power consumption obtained by dividing a luminescence flux by a power consumption.
 - 15. An organic luminescence device according to Claim 1, wherein phosphorescence is emitted from the luminescence layer by applying a voltage between the electrodes.
 - 16. A picture display apparatus, comprising an organic luminescence device according to Claim 1, and a drive-circuit for supplying display data.

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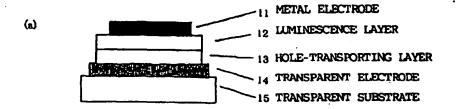
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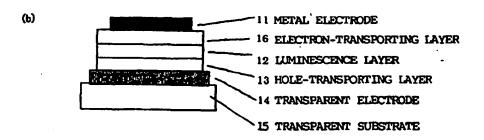
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FIG. 1





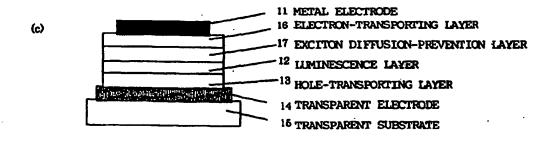
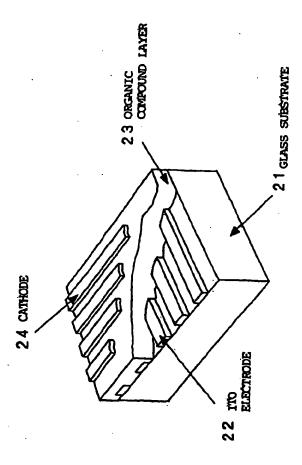


FIG. 2





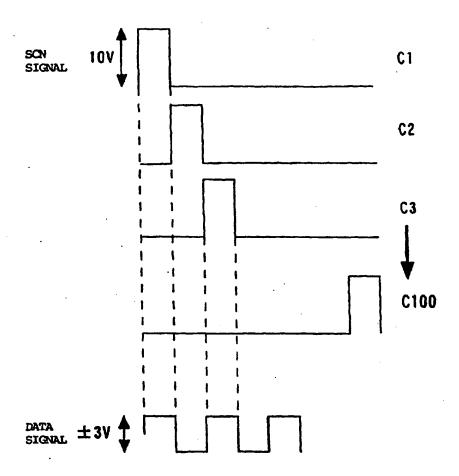


FIG. 4

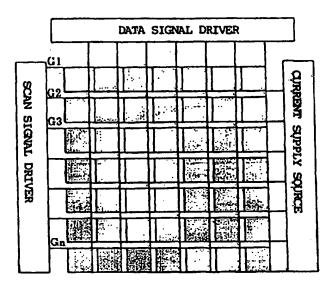


FIG. 5

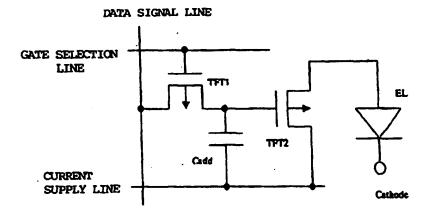
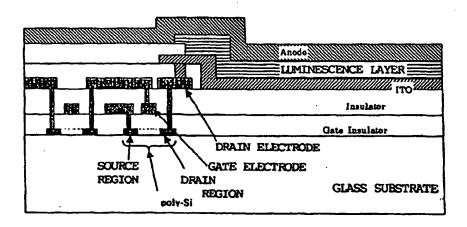


FIG. 6



INTERNATIONAL SEARCH REPORT

International application No.

PCT/JP01/10477

	SIFICATION OF SUBJECT MATTER	22 15/00 19/00				
1110	Int.Cl ⁷ H05B 33/14, C09K 11/06, C07F 15/00, 19/00					
According t	According to International Patent Classification (IPC) or to both national classification and IPC					
B. FIELDS SEARCHED						
Minimum documentation searched (classification system followed by classification symbols) Int.Cl ⁷ H05B 33/00-33/28, C09K 11/06, C07F 15/00, 19/00						
Jits	Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Jitsuyo Shinan Koho 1922-1996 Toroku Jitsuyo Shinan Koho 1994-2002 Kokai Jitsuyo Shinan Koho 1971-2002 Jitsuyo Shinan Toroku Koho 1996-2002					
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) CA (STN), JICST FILE (JOIS)						
C. DOCU	C. DOCUMENTS CONSIDERED TO BE RELEVANT					
Category*	Citation of document, with indication, where ap	opropriate, of the relevant passages	Relevant to claim No.			
PΧ	US 2001/19782 A1 (IGARASHI, Tat 06 September, 2001 (06.09.2001) Full text		1,4,6-10, 15,16			
· !	& JP 2001-247859 A Full text & JP 2001-345183 A Full text					
A	BALDO et al., "Highly Efficient from Organic Electroluminescent Nature, (1998), Vol.395, pages	Devices",	1-16			
	TSUTSUI, Tetsuo et al., "High (Organic Light-Emitting Devices as a Triplet Emissive Center", (1999), Vol.38, Part 2, No.12(E	1-16				
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	ent published prior to the international filing date but later priority date claimed	"&" document member of the same patent				
Date of the actual completion of the international search 04 February, 2002 (04.02.02)		Date of mailing of the international search report 19 February, 2002 (19.02.02)				
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